

# National Testing Agency

**Question Paper Name:** 5271Fundamentals of Chemical Engineering30th June 2019 Shift 2  
**Subject Name:** Fundamentals of Chemical Engineering  
**Creation Date:** 2019-06-30 18:16:53  
**Duration:** 180  
**Total Marks:** 100  
**Display Marks:** Yes

## Fundamentals of Chemical Engineering

**Group Number :** 1  
**Group Id :** 489994195  
**Group Maximum Duration :** 0  
**Group Minimum Duration :** 120  
**Revisit allowed for view? :** No  
**Revisit allowed for edit? :** No  
**Break time:** 0  
**Group Marks:** 100

## Fundamentals of Chemical Engineering

**Section Id :** 489994251  
**Section Number :** 1  
**Section type :** Online  
**Mandatory or Optional:** Mandatory  
**Number of Questions:** 72  
**Number of Questions to be attempted:** 72  
**Section Marks:** 100  
**Display Number Panel:** Yes  
**Group All Questions:** No

**Sub-Section Number:** 1  
**Sub-Section Id:** 489994269  
**Question Shuffling Allowed :** Yes

**Question Number : 1 Question Type : MCQ Option Shuffling : No Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical**

**Correct Marks : 1 Wrong Marks : 0**

The dimension of 'Thermal conductivity' is

a)  $\frac{ML\theta^{-3}}{T^{-1}}$     b)  $\frac{MTL^{-3}}{\theta^{-1}}$     c)  $\frac{MLT^{-3}}{\theta^{-1}}$     d)  $\frac{\theta LT^{-3}}{M^{-1}}$

**Options :**

1. 1
2. 2
3. 3
4. 4

Question Number : 2 Question Type : MCQ Option Shuffling : No Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

Correct Marks : 1 Wrong Marks : 0

The maximum adiabatic flame temperature in air as compared to that in pure oxygen is

- a) Much lower
- b) Much higher
- c) same
- d) either lower or higher, depends on the type of fuel

Options :

- 1. 1
- 2. 2
- 3. 3
- 4. 4

Question Number : 3 Question Type : MCQ Option Shuffling : No Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

Correct Marks : 1 Wrong Marks : 0

Degree of freedom in a mass balance problem is  $< 1$ . This indicates

- a) The problem can be solved
- b) More unknowns than independent equations
- c) Variables are underspecified
- d) Flowchart is incompletely labelled

Options :

- 1. 1
- 2. 2
- 3. 3
- 4. 4

Question Number : 4 Question Type : MCQ Option Shuffling : No Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

Correct Marks : 1 Wrong Marks : 0

When a flowsheet is scaled, which of the following changes?

- a) Density of streams
- b) Mass fraction of components in a stream
- c) Molar flowrate of components in a stream
- d) Temperature of streams

Options :

- 1. 1
- 2. 2
- 3. 3
- 4. 4

Question Number : 5 Question Type : MCQ Option Shuffling : No Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

Correct Marks : 1 Wrong Marks : 0

Air (21 mole % O<sub>2</sub> and 79 mole % N<sub>2</sub>) and 100 mol of HCl are supplied into a reactor for the manufacture of chlorine. Amount of air supplied is such that oxygen is in 35 % excess. The degree of freedom for this process is

- a) 0
- b) 1
- c) 2
- d) 3

**Options :**

- 1. 1
- 2. 2
- 3. 3
- 4. 4

**Question Number : 6 Question Type : MCQ Option Shuffling : No Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical**

**Correct Marks : 1 Wrong Marks : 0**

Number of intensive variables that can be specified independently for a closed system containing a vapour-liquid mixture of benzene-toluene at equilibrium is

- a) 3
- b) 2
- c) 1
- d) 0

**Options :**

- 1. 1
- 2. 2
- 3. 3
- 4. 4

**Question Number : 7 Question Type : MCQ Option Shuffling : No Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical**

**Correct Marks : 1 Wrong Marks : 0**

200 kg of an aqueous solution of KNO<sub>3</sub> (60 mass %) at 80°C is fed to a crystallizer where the solution is cooled to 40°C. The saturation concentration of KNO<sub>3</sub> at 40°C is 38.6 mass %. The degree of freedom for this process is

- a) 0
- b) 1
- c) 2
- d) 3

**Options :**

- 1. 1
- 2. 2
- 3. 3
- 4. 4

**Question Number : 8 Question Type : MCQ Option Shuffling : No Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical**

**Correct Marks : 1 Wrong Marks : 0**

An approximated equation may be used to estimate the dew-point temperature for known dry bulb temperature and humidity is as

$$\frac{4030 * (T + 253)}{4030 - (T + 253) * \ln(x)} - 253$$

where x is,

- |                      |                     |
|----------------------|---------------------|
| a. Absolute humidity | b. Molal humidity   |
| b. Relative humidity | d. Percent humidity |

Options :

1. 1
2. 2
3. 3
4. 4

Question Number : 9 Question Type : MCQ Option Shuffling : No Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

Correct Marks : 1 Wrong Marks : 0

The concept of wet bulb temperature is based on the equilibrium between the rates of energy transfer to the \_\_\_\_\_ and the evaporation of \_\_\_\_\_

- a) Water, vapour
- b) Bulb, water
- c) Bulb, vapour
- d) Vapour, water

Options :

1. 1
2. 2
3. 3
4. 4

Question Number : 10 Question Type : MCQ Option Shuffling : No Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

Correct Marks : 1 Wrong Marks : 0

Heat transfer coefficient (h) for liquids increases with

- |                               |                           |
|-------------------------------|---------------------------|
| a) increasing temperature     | b) decreasing temperature |
| c) decreasing Reynolds number | d) none of these          |

Options :

1. 1
2. 2
3. 3
4. 4

Question Number : 11 Question Type : MCQ Option Shuffling : No Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

Correct Marks : 1 Wrong Marks : 0



Which boiling regime is the most economical and desirable in a boiling equipment?

- (a): film boiling
- (b): free convective boiling
- (c): nucleate boiling
- (d): transition boiling

Options :

- 1. 1
- 2. 2
- 3. 3
- 4. 4

Question Number : 15 Question Type : MCQ Option Shuffling : No Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

Correct Marks : 1 Wrong Marks : 0

Heat transfer coefficient ( $h$ ) in the nucleate boiling varies with temperature difference between the heating element and the liquid ( $\Delta t$ ) as:

- (a):  $h \propto \Delta T^2$
- (b):  $h \propto \Delta T^{1.25}$
- (c):  $h \propto \Delta T^1$
- (d):  $h \propto \Delta T^{0.5}$

Options :

- 1. 1
- 2. 2
- 3. 3
- 4. 4

Question Number : 16 Question Type : MCQ Option Shuffling : No Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

Correct Marks : 1 Wrong Marks : 0

In a counter current 1-1 shell and tube heat exchanger, the following are the inlet and outlet temperatures:

- (a) Hot fluid inlet temperature=120 °C
- (b) Hot fluid outlet temperature=100 °C
- (c) Cold fluid outlet temperature=70 °C
- (d) Cold fluid inlet temperature=50 °C

The mean temperature difference to be used is.....

Options :

- 1. 1
- 2. 2
- 3. 3
- 4. 4

Question Number : 17 Question Type : MCQ Option Shuffling : No Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

Correct Marks : 1 Wrong Marks : 0

Select the correct option:

A. The log mean temperature difference (LMTD) for counter flow and parallel flow can be same when any one of the fluids passes through the heat exchanger at a constant temperature

B. Shell side pressure drop in a shell and tube heat exchanger does not depend upon the shell diameter

- (a) Both the statements are true
- (b) Both the statements are false
- (c) Only statement in (A) is true
- (d) Only statement in (B) is true

Options :

- 1. 1
- 2. 2
- 3. 3
- 4. 4

Question Number : 18 Question Type : MCQ Option Shuffling : No Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

Correct Marks : 1 Wrong Marks : 0

Increasing the liquor level in the evaporator results in

- a) Decreased capacity
- b) Increase in liquor film coefficient
- c) Decreased effect of hydrostatic head
- d) Increased true temperature drop

Options :

- 1. 1
- 2. 2
- 3. 3
- 4. 4

Question Number : 19 Question Type : MCQ Option Shuffling : No Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

Correct Marks : 1 Wrong Marks : 0

Economy of an evaporator is influenced by

- a) Steam pressure
- b) Temperature of the feed
- c) Number of effect
- d) Both b and c

Options :

- 1. 1
- 2. 2
- 3. 3
- 4. 4

Question Number : 20 Question Type : MCQ Option Shuffling : No Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

Correct Marks : 1 Wrong Marks : 0

The operating line for an absorber is curved when plotted in terms of

- a) Mole ratio
- b) Mole fraction
- c) Mass ratio
- d) None of the above.

Options :

- 1. 1
- 2. 2
- 3. 3
- 4. 4

Question Number : 21 Question Type : MCQ Option Shuffling : No Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

Correct Marks : 1 Wrong Marks : 0

In case of a stripper

- a) The operating line will always lie above the equilibrium curve
- b) The operating line will always lie below the equilibrium curve
- c) There is no fixed rule for relative position of operating line and equilibrium curve
- d) None of these

Options :

- 1. 1
- 2. 2
- 3. 3
- 4. 4

Question Number : 22 Question Type : MCQ Option Shuffling : No Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

Correct Marks : 1 Wrong Marks : 0

What is the physical significance of the absorption factor A?

- a) It is the ratio of the slopes of the equilibrium line and the operating line.
- b) It is ratio of the slopes of the operating line and the equilibrium line.
- c) It is fractional absorption of the feed.
- d) None of these

Options :

- 1. 1
- 2. 2
- 3. 3
- 4. 4

Question Number : 23 Question Type : MCQ Option Shuffling : No Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

Correct Marks : 1 Wrong Marks : 0



The minimum liquid rate to be used in an absorber corresponds to an operating line

- a) Of slope = 1
- b) Of slope = 0.1
- c) Tangential to the equilibrium curve
- d) None of these

Options :

- 1. 1
- 2. 2
- 3. 3
- 4. 4

Question Number : 24 Question Type : MCQ Option Shuffling : No Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

Correct Marks : 1 Wrong Marks : 0

If the gas phases composition of a component A is 0.65 and its relative volatility is 2.

Find the liquid phase composition.

- a) 0.48
- b) 0.58
- c) 0.68
- d) 0.78

Options :

- 1. 1
- 2. 2
- 3. 3
- 4. 4

Question Number : 25 Question Type : MCQ Option Shuffling : No Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

Correct Marks : 1 Wrong Marks : 0

The column used for differential distillation is \_\_\_\_\_

- a) Still
- b) Differential column
- c) Batch column
- d) None of the mentioned

Options :

- 1. 1
- 2. 2
- 3. 3
- 4. 4

Question Number : 26 Question Type : MCQ Option Shuffling : No Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

Correct Marks : 1 Wrong Marks : 0

If the feed enters the distillation column at its dew point, then the slope of the Flash vaporization operating line is,

- a) Zero
- b) Infinity
- c) 1
- d) Between 0 and 1

Options :

- 1. 1
- 2. 2

3. 3

4. 4

**Question Number : 27 Question Type : MCQ Option Shuffling : No Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical**

**Correct Marks : 1 Wrong Marks : 0**

If there is no reflux to a fractionating column then

- a) Large condenser size is needed
- b) Less reboiler size needed
- c) Minimum number of trays
- d) None of the mentioned

**Options :**

1. 1

2. 2

3. 3

4. 4

**Question Number : 28 Question Type : MCQ Option Shuffling : No Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical**

**Correct Marks : 1 Wrong Marks : 0**

McCabe –Thiele method excludes \_\_\_\_\_ information.

- a) Enthalpy
- b) Entropy
- c) Flow rate
- d) Number of theoretical stages

**Options :**

1. 1

2. 2

3. 3

4. 4

**Question Number : 29 Question Type : MCQ Option Shuffling : No Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical**

**Correct Marks : 1 Wrong Marks : 0**

If the feed is saturated vapor, the slope of q-line is,

- a) 1
- b) infinity
- c) Zero
- d) None

**Options :**

1. 1

2. 2

3. 3

4. 4

**Question Number : 30 Question Type : MCQ Option Shuffling : No Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical**

**Correct Marks : 1 Wrong Marks : 0**

The composition of substances in the extraction are represented by \_\_\_\_\_

- a) Rectangle
- b) Isosceles triangle
- c) Equilateral triangle
- d) None of the mentioned

Options :

- 1. 1
- 2. 2
- 3. 3
- 4. 4

Question Number : 31 Question Type : MCQ Option Shuffling : No Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

Correct Marks : 1 Wrong Marks : 0

Number of possible operating points in a nonisothermal CSTR is,

- (a) 1
- (b) 2
- (c) 3
- (d) Any one of the above options

Options :

- 1. 1
- 2. 2
- 3. 3
- 4. 4

Question Number : 32 Question Type : MCQ Option Shuffling : No Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

Correct Marks : 1 Wrong Marks : 0

The nature of the heat removal curve in a nonisothermal CSTR is

- a) Nearly Linear
- b) Hyperbolic
- c) Sigmoidal
- d) None of the above options

Options :

- 1. 1
- 2. 2
- 3. 3
- 4. 4

Question Number : 33 Question Type : MCQ Option Shuffling : No Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

Correct Marks : 1 Wrong Marks : 0

For stability in a nonisothermal CSTR Eigen values of the Jacobian for coefficient matrix of the conservation equations must have

- (a) Positive real parts
- (b) Negative real parts
- (c) Negative imaginary parts
- (d) Positive imaginary parts

Options :

- 1. 1
- 2. 2

3. 3

4. 4

**Question Number : 34 Question Type : MCQ Option Shuffling : No Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical**

**Correct Marks : 1 Wrong Marks : 0**

Ignition temperature in an isothermal CSTR is defined as:

- (a) The reactor temperature at which the reactor steady state shoots from the lower steady state to upper steady state
- (b) The heating/cooling medium temperature at which the reactor steady state shoots from the lower steady state to upper steady state
- (c) The feed temperature at which the reactor steady state shoots from the lower steady state to upper steady state
- (d) None of the above options

**Options :**

1. 1

2. 2

3. 3

4. 4

**Question Number : 35 Question Type : MCQ Option Shuffling : No Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical**

**Correct Marks : 1 Wrong Marks : 0**

Unreacted shrinking core model is more appropriate for application when

- (a) Both the solid reactant and solid product are impervious
- (b) The solid reactant is porous and solid product is impervious
- (c) The solid reactant is impervious and solid product is porous
- (d) None of the above options

**Options :**

1. 1

2. 2

3. 3

4. 4

**Question Number : 36 Question Type : MCQ Option Shuffling : No Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical**

**Correct Marks : 1 Wrong Marks : 0**

In random pore model for gas-solid noncatalytic reactions

- (a) The pore structure evolution is incorporated as a known function of time
- (b) The pore structure evolution is incorporated as constant
- (c) The pore structure evolution is incorporated as a function of conversion
- (d) None of the above

**Options :**

1. 1

2. 2

3. 3

4. 4

**Question Number : 37 Question Type : MCQ Option Shuffling : No Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical**

**Correct Marks : 1 Wrong Marks : 0**

When a gas-solid noncatalytic reaction following unreacted shrinking core model is controlled by product layer diffusion resistance, the reaction completion time?

- (a)  $\tau \propto R^{1.5}$
- (b)  $\tau \propto R^2$
- (c)  $\tau \propto R$
- (d) None of the above options

Options :

- 1. 1
- 2. 2
- 3. 3
- 4. 4

Question Number : 38 Question Type : MCQ Option Shuffling : No Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

Correct Marks : 1 Wrong Marks : 0

In a gas-solid noncatalytic reactor if the solid particles are of different but unchanging size and in plug flow and the reaction is controlled by chemical reaction resistance, the design equation is

$$(a) \quad 1 - \overline{X}_B = \sum_{R_i(t_p - \tau)}^{R_m} \left[ 1 - \frac{t_p}{\tau(R_i)} \right]^2 \frac{F(R_i)}{F}$$

$$(b) \quad 1 - \overline{X}_B = \sum_{R_i(t_p - \tau)}^{R_m} \left[ 1 - \frac{t_p}{\tau(R_i)} \right] \frac{F(R_i)}{F}$$

$$(c) \quad 1 - \overline{X}_B = \sum_{R_i(t_p - \tau)}^{R_m} \left[ 1 - \frac{t_p}{\tau(R_i)} \right]^{1/3} \frac{F(R_i)}{F}$$

- (d) None of the above options

Options :

- 1. 1
- 2. 2
- 3. 3
- 4. 4

Question Number : 39 Question Type : MCQ Option Shuffling : No Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

Correct Marks : 1 Wrong Marks : 0

Thrust on a pipe bend is obtained using

- a) Microscopic mass balance
- b) Microscopic linear momentum balance
- c) Macroscopic linear momentum balance
- d) Macroscopic kinetic energy balance

Options :

- 1. 1
- 2. 2
- 3. 3

4. 4

Question Number : 40 Question Type : MCQ Option Shuffling : No Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

Correct Marks : 1 Wrong Marks : 0

Choose the odd-man out

- a) Mass balance
- b) Energy balance
- c) Momentum balance
- d) Entropy balance

Options :

- 1. 1
- 2. 2
- 3. 3
- 4. 4

Question Number : 41 Question Type : MCQ Option Shuffling : No Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

Correct Marks : 1 Wrong Marks : 0

Concentration profile is obtained by solving

- a) Differential linear momentum balance
- b) Differential energy balance
- c) Differential species mass balance
- d) Differential total mass balance

Options :

- 1. 1
- 2. 2
- 3. 3
- 4. 4

Question Number : 42 Question Type : MCQ Option Shuffling : No Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

Correct Marks : 1 Wrong Marks : 0

Viscous stress can also be interpreted as

- a) Potential energy
- b) Kinetic energy
- c) Momentum flux
- d) Pressure

Options :

- 1. 1
- 2. 2
- 3. 3
- 4. 4

Question Number : 43 Question Type : MCQ Option Shuffling : No Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

Correct Marks : 1 Wrong Marks : 0

Equimolar counter diffusion between A and B assumes

- a) A and B must react
- b) A and B must not react
- c) A and B must repeal each other
- d) Diameter of A and B must be equal

Options :

- 1. 1
- 2. 2
- 3. 3
- 4. 4

Question Number : 44 Question Type : MCQ Option Shuffling : No Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

Correct Marks : 1 Wrong Marks : 0

Species continuity equation is applicable only for

- a) Constant density fluid
- b) Only kind of fluid
- c) Gaseous fluid
- d) Gases under very high vacuum

Options :

- 1. 1
- 2. 2
- 3. 3
- 4. 4

Question Number : 45 Question Type : MCQ Option Shuffling : No Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

Correct Marks : 1 Wrong Marks : 0

In a mixture of n number of species, number of species continuity equation expressed in terms of mole fraction is

- a)  $n+1$
- b)  $n^2$
- c)  $n-1$
- d)  $2n$

Options :

- 1. 1
- 2. 2
- 3. 3
- 4. 4

Question Number : 46 Question Type : MCQ Option Shuffling : No Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

Correct Marks : 1 Wrong Marks : 0

Navier-Stokes equation contains following term

- a) Accumulation and convection terms
- b) Convection and diffusion terms
- c) Convection, diffusion, accumulation terms
- d) Convection, diffusion, accumulation and source terms

Options :

- 1. 1
- 2. 2
- 3. 3
- 4. 4

Question Number : 47 Question Type : MCQ Option Shuffling : No Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

Correct Marks : 1 Wrong Marks : 0

Consider flow of a binary gas mixture in a cylindrical tube. The tube walls are selectively permeable to only one species. The Peclet number and Schmidt number are  $Pe$  and  $Sc$ , respectively. Axial diffusion can be neglected in modeling of concentration profile in the tube when:

- a.  $Pe \gg 1$
- b.  $Pe \ll 1$
- c.  $Sc \gg 1$
- d.  $Sc \ll 1$

Options :

- 1. 1
- 2. 2
- 3. 3
- 4. 4

Question Number : 48 Question Type : MCQ Option Shuffling : No Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

Correct Marks : 1 Wrong Marks : 0

The total molecular energy flux for a multicomponent mixture consists of heat transport by conduction and an additional term resulting on account of:

- a. difference in partial specific entropy of constituent species
- b. difference in density of constituent species
- c. difference in partial specific enthalpy of constituent species
- d. difference in viscosity of constituent species

Options :

- 1. 1
- 2. 2
- 3. 3
- 4. 4

Question Number : 49 Question Type : MCQ Option Shuffling : No Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

Correct Marks : 1 Wrong Marks : 0



Consider a multicomponent mixture, for which diffusion-thermo effect is not important, i.e., it can be ignored. Pick the correct statement:

- The total molecular energy flux for given multicomponent mixture will be equal to heat flux by conduction when all diffusional mass fluxes are zero.
- The total molecular energy flux for given multicomponent mixture will be always equal to heat flux by conduction.
- The total molecular energy flux for given multicomponent mixture will be equal to heat flux by conduction when the mixture can be modeled as an ideal solution.
- The total molecular energy flux for given multicomponent mixture will be equal to heat flux by conduction when the diffusional mass fluxes can be modeled by Maxwell-Stefan equations.

Options :

- 1
- 2
- 3
- 4

Question Number : 50 Question Type : MCQ Option Shuffling : No Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

Correct Marks : 1 Wrong Marks : 0

The representative volume element (RVE) is the size-scale over which the transport processes (mass, momentum, energy) can be modeled by the continuum approximation. Pick the correct option for relationship of RVE with molecular volume ( $V_m$ ) and volume of system being modeled ( $V$ ):

- $RVE \gg V_m, V$
- $RVE \ll V_m, V$
- $V_m \ll RVE \ll V$
- $RVE = V_m$

Options :

- 1
- 2
- 3
- 4

Sub-Section Number: 2  
Sub-Section Id: 489994270  
Question Shuffling Allowed : Yes

Question Number : 51 Question Type : MCQ Option Shuffling : No Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

Correct Marks : 2 Wrong Marks : 0

Grashoff number (G) is defined as

- $(\text{Inertia force} \times \text{buoyancy force})/(\text{viscous force})^2$
- $(\text{Inertia force} \times \text{viscous force})/(\text{buoyancy force})^2$
- $(\text{Buoyancy force} \times \text{viscous force})/(\text{inertia force})^2$
- None of the above

Options :

- 1

2. 2
3. 3
4. 4

Question Number : 52 Question Type : MCQ Option Shuffling : No Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

Correct Marks : 2 Wrong Marks : 0

Fresh air (4 mole % water vapour) is mixed with recycled dehumidified air and passed into an air cooler in which 1 mol of water vapour condenses out as liquid. Part of the dehumidified air leaving the cooler is recycled and the balance is delivered to a room. 100 mol of dehumidified air (1.7 mole % water vapour) is delivered to the room. The degree of freedom for this process is

- a) -1
- b) 0
- c) 1
- d) 2

Options :

1. 1
2. 2
3. 3
4. 4

Question Number : 53 Question Type : MCQ Option Shuffling : No Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

Correct Marks : 2 Wrong Marks : 0

100 kmol of equimolar mixture containing ethylene and oxygen is fed to a reactor in which ethylene oxide is formed. The fractional conversion of ethylene is 30%. The moles of  $O_2$  leaving the reactor is

- a) 42.5
- b) 35
- c) 15
- d) 25

Options :

1. 1
2. 2
3. 3
4. 4

Question Number : 54 Question Type : MCQ Option Shuffling : No Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

Correct Marks : 2 Wrong Marks : 0

Calculate the heat transfer to the atmosphere per second from a circular 5cm in diameter and 100 m long carrying steam at an average temperature of  $120^\circ\text{C}$  if surroundings are at  $20^\circ\text{C}$ . The heat transfer can be estimated from the relation  $Q = h \cdot A \cdot \Delta T$ , where A is the surface area of the pipe and  $\Delta T$  is temperature difference between surface of the pipe and ambient condition.

- (a) 7854 J/s
- (b) 5800 J/s
- (c) 4800 J/s
- (d) 9700 J/s.

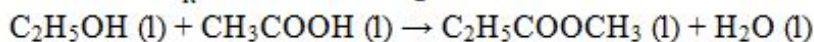
Options :

- 1
- 2
- 3
- 4

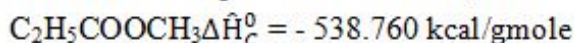
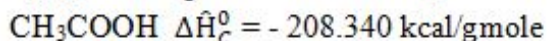
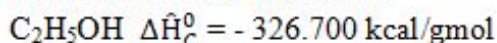
Question Number : 55 Question Type : MCQ Option Shuffling : No Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

Correct Marks : 2 Wrong Marks : 0

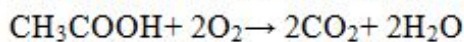
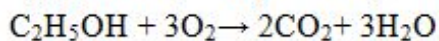
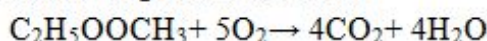
Calculate  $\Delta \hat{H}_R^0$  for the following reaction:



Heat of combustion data are:



The following reactions occur:



- (a) 420.4 kcal      (b) 657.12 kcal      (c) 3.720 kcal      (d) 212.06 kcal

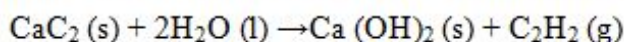
Options :

- 1
- 2
- 3
- 4

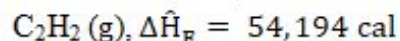
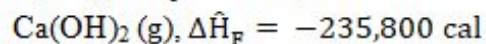
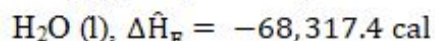
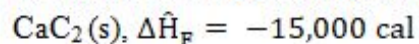
Question Number : 56 Question Type : MCQ Option Shuffling : No Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

Correct Marks : 2 Wrong Marks : 0

Calculate the standard heat of reaction for the following reaction:



Standard heat of formation data is:



- (a) -333.24 kcal/gmole      (b) -98.289 kcal/gmole  
(c) -181.606 kcal/gmole      (d) -29.971 kcal/gmole

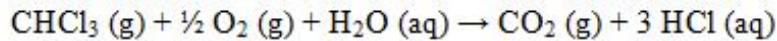
Options :

- 1
- 2
- 3
- 4

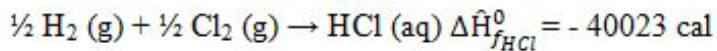
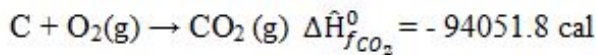
Question Number : 57 Question Type : MCQ Option Shuffling : No Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

Correct Marks : 2 Wrong Marks : 0

Calculate the heat of formation of chloroform, if the standard heat of combustion of chloroform is  $\Delta \hat{H}_C^0 = -121,800$  cal/mole as per the following reaction:



Data provided:  $\text{H}_2(\text{g}) + \frac{1}{2} \text{O}_2(\text{g}) \rightarrow \text{H}_2\text{O}(\text{l}) \Delta \hat{H}_{f_{\text{H}_2\text{O}}}^0 = -68317.4$  cal



- (a) - 24 kcal/gmole  
(c) - 79.5 kcal/gmole
- (b) - 24 cal/gmole  
(d) - 160.6337 cal/gmole

Options :

- 1
- 2
- 3
- 4

Question Number : 58 Question Type : MCQ Option Shuffling : No Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

Correct Marks : 2 Wrong Marks : 0

For a steel pipe heated to 373 K and kept in 300 K atmosphere, Biot number is the ratio of

- (a) Convection heat transfer at the boundary of steel pipe/conduction inside the steel pipe
- (b) Convection heat transfer in the steel pipe/conduction inside the steel pipe
- (c) Convection heat transfer in the steel pipe/conduction at the boundary of steel pipe
- (d) Convection heat transfer at the boundary of steel pipe/ conduction at the boundary of steel pipe

Options :

- 1
- 2
- 3
- 4

Question Number : 59 Question Type : MCQ Option Shuffling : No Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

Correct Marks : 2 Wrong Marks : 0

The bottom of a metal pan, 0.3 m in diameter, is kept at 118 °C by an electric heating system. What is the evaporation rate of water (in kg/hr) if the heat flux is 836 kW/m<sup>2</sup> and the latent heat of water is 2257 kJ/kg?

- (A): 1333
- (B): 94
- (C): 0.37
- (D): 0.026

Options :

- 1
- 2
- 3
- 4

Question Number : 60 Question Type : MCQ Option Shuffling : No Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

Correct Marks : 2 Wrong Marks : 0

In a film boiling experiment, a horizontal cylindrical heating element is used. How does the heat transfer coefficient changes when the diameter of the cylinder is doubled with other variable being held unchanged?

- (a):  $h$  increases by a factor of 2
- (b):  $h$  increases by a factor of  $2^{1/4}$
- (c):  $h$  increases by a factor of  $2^{3/4}$
- (d):  $h$  does not change

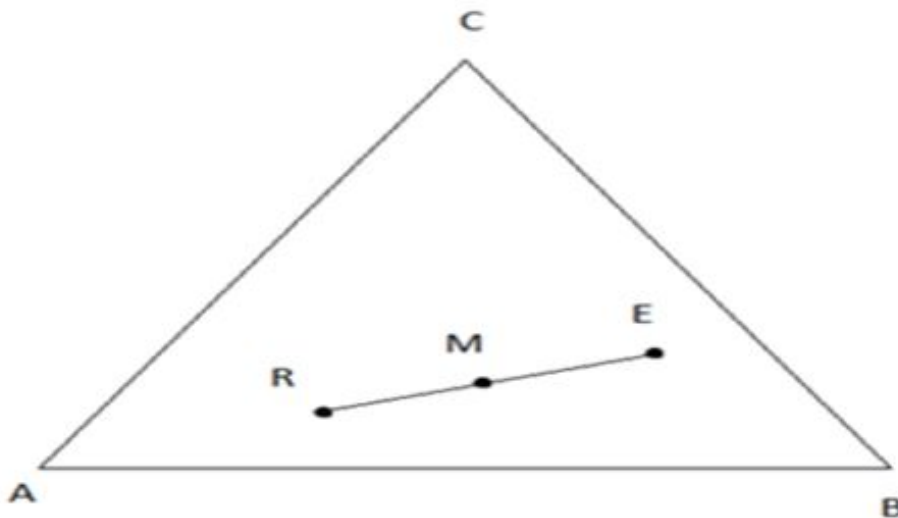
Options :

- 1. 1
- 2. 2
- 3. 3
- 4. 4

Question Number : 61 Question Type : MCQ Option Shuffling : No Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

Correct Marks : 2 Wrong Marks : 0

10 kg Raffinate(R) of 20% solute concentration is mixed with 4 kg fresh extracting solvent (E) to separate the solute from raffinate. The solute concentration in mixture ( $x_M$ ) will be,



- a) 1.43
- b) 0.14
- c) 0.2
- d) none

Options :

- 1. 1
- 2. 2
- 3. 3
- 4. 4

Question Number : 62 Question Type : MCQ Option Shuffling : No Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

Correct Marks : 2 Wrong Marks : 0

In a batch adsorption process, 5 g of fresh adsorbent is used to treat 1 liter of an aqueous phenol solution. The initial phenol concentration is 100 mg/liter. The equilibrium relation is given by  $q^* = 1.3 C$  where  $q^*$  is the amount of phenol adsorbed in mg of phenol per gram of adsorbent; and  $C$  is the concentration of phenol in mg/liter in the aqueous solution. When equilibrium is attained between the adsorbent and the solution, the concentration of phenol in the solution, rounded to 1 decimal place, in mg/liter is

- |     |          |     |           |
|-----|----------|-----|-----------|
| [a] | 130 -132 | [b] | 13.0-13.5 |
| [c] | 0.1-0.2  | [d] | 5.1-5.3   |

Options :

1. 1
2. 2
3. 3
4. 4

Question Number : 63 Question Type : MCQ Option Shuffling : No Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

Correct Marks : 2 Wrong Marks : 0

The relationship between average velocity and maximum velocity in case of flow between two parallel plates is

- |                             |                            |
|-----------------------------|----------------------------|
| (a) $U_{max} = U_{av}$      | (b) $U_{max} = 0.5 U_{av}$ |
| (c) $U_{max} = 0.75 U_{av}$ | (d) $U_{max} = 1.5 U_{av}$ |

Options :

1. 1
2. 2
3. 3
4. 4

Question Number : 64 Question Type : MCQ Option Shuffling : No Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

Correct Marks : 2 Wrong Marks : 0

Oil flows between two parallel plates, one of which is at rest and the other moves at a velocity,  $U$ . If the pressure is decreasing in the direction of the flow at the rate of 5 Pa/m, dynamic viscosity is 0.05 kg/ms, the spacing of the horizontal plate is 0.04 m and the volumetric flow  $Q$  per unit width is 0.02 m<sup>2</sup>/s, the value of  $U$  is

- |                |               |
|----------------|---------------|
| (a) 10.067 m/s | (b) 1.027 m/s |
| (c) 0.097 m/s  | (d) 0.97 m/s  |

Options :

1. 1
2. 2
3. 3
4. 4

Question Number : 65 Question Type : MCQ Option Shuffling : No Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

Correct Marks : 2 Wrong Marks : 0

Consider a catalytic tubular reactor where an irreversible reaction  $A \rightarrow B$  takes place under isothermal conditions. A feed stream containing species  $A$  enters the reactor and a stream containing both  $A$ ,  $B$  is obtained at the outlet. The viscosity of fluid stream is dependent on concentration of species  $B$ . Pick the correct statement regarding governing equations for steady-state concentration profiles,  $C_i(r, z)$ , and velocity profile,  $v(r, z)$ , in the reactor:

- Solution for  $C_i$  can be obtained independently from  $\mu$ , however, equation for  $v$  is coupled to  $C_i$ .
- Solution for  $v$  can be obtained independently from  $C_i$ , however, equation for  $C_i$  is coupled to  $v$ .
- Equations for  $C_i$  and  $v$  can be solved independently from each-other.
- Equations for  $C_i$  and  $v$  are coupled to each-other and need to be solved simultaneously.

Options :

- 1
- 2
- 3
- 4

Question Number : 66 Question Type : MCQ Option Shuffling : No Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

Correct Marks : 2 Wrong Marks : 0

Consider 1D energy transport (by conduction only) in  $x$ -direction across a long, thin slab of thickness  $W$ . The time dependence of temperature at the upstream  $yz$ -surface of slab is given by  $T_0 \exp(-k_T t)$  while the temperature at the downstream surface is held constant. The thermal diffusion coefficient in the slab is given by  $\alpha_T$ . Pick the condition at which a pseudo steady-state analysis can be used to model the temperature profile in the slab:

- $1/k_T \gg W^2 / \alpha_T$
- $1/k_T \gg \alpha_T / W^2$
- $1/k_T \ll W^2 / \alpha_T$
- $1/k_T \ll \alpha_T / W^2$

Options :

- 1
- 2
- 3
- 4

Sub-Section Number: 3  
Sub-Section Id: 489994271  
Question Shuffling Allowed : Yes

Question Number : 67 Question Type : MCQ Option Shuffling : No Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

Correct Marks : 3 Wrong Marks : 0

If the capillary rise 'h' is found to be influenced by the tube diameter D, density  $\rho$ , gravitational acceleration g and surface tension  $\sigma$ . The dimensional groups predicted can be

a)  $\frac{g}{\sigma^2 D \rho} \frac{h}{D}$

b)  $\frac{D}{\sigma^2 g \rho} \frac{D}{h}$

c)  $\frac{\sigma}{D^2 g \rho} \frac{h}{D}$

d) None of the above

Options :

1. 1
2. 2
3. 3
4. 4

Question Number : 68 Question Type : MCQ Option Shuffling : No Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

Correct Marks : 3 Wrong Marks : 0

2000 lph of a benzene-toluene mixture (density  $872 \text{ kg/m}^3$ ) of 45 mass % benzene enters a distillation column. The top product is 95 mole % benzene and 8 % of benzene in the feed is in the bottom product. Mass flowrate of toluene (kg/h) leaving through the bottom product is

- a) 915
- b) 716
- c) 630
- d) 744

Options :

1. 1
2. 2
3. 3
4. 4

Question Number : 69 Question Type : MCQ Option Shuffling : No Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

Correct Marks : 3 Wrong Marks : 0

Dehydrogenation of propane takes place in a reactor fed with propane and propylene. Moles of propane, propylene and hydrogen leaving the reactor are 900, 99.75 and 95. The single pass conversion of propane is

- a) 9.5%
- b) 12.8%
- c) 6.3%
- d) 18.9%

Options :

1. 1
2. 2
3. 3
4. 4



Question Number : 70 Question Type : MCQ Option Shuffling : No Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

Correct Marks : 3 Wrong Marks : 0

Air (21 mole % O<sub>2</sub> and 79 mole % N<sub>2</sub>) and 100 mol of HCl are supplied into a reactor for the manufacture of chlorine. Amount of air supplied is such that oxygen is in 35 % excess and 85 % conversion of HCl is achieved. The moles of O<sub>2</sub> leaving the reactor is

- a) 12.5
- b) 42.5
- c) 15
- d) 30

Options :

- 1. 1
- 2. 2
- 3. 3
- 4. 4

Question Number : 71 Question Type : MCQ Option Shuffling : No Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

Correct Marks : 3 Wrong Marks : 0

A solution is ultrafiltered with feed concentration  $c_0$  and permeates concentration  $c_p$ . If  $k$  is the mass transfer coefficient and  $v_w$  is the permeate flux, using film theory, obtain an expression of observed retention in terms of real retention ( $R_r$ ),  $v_w$  and  $k$ .

(a)  $\ln \left( \frac{R_0}{1-R_0} \right) = \ln \left( \frac{R_r}{1-R_r} \right) - \frac{v_w}{k}$

(b)  $\ln \left( \frac{R_0}{1+R_0} \right) = \ln \left( \frac{R_r}{1+R_r} \right) - \frac{v_w}{k}$

(c)  $\ln \left( \frac{R_0}{1+R_0} \right) = \ln \left( \frac{R_r}{1-R_r} \right) + \frac{v_w}{k}$

(d)  $\ln \left( \frac{R_0}{1-R_0} \right) = \ln \left( \frac{R_r}{1+R_r} \right) + \frac{v_w}{k}$

Options :

- 1. 1
- 2. 2
- 3. 3
- 4. 4

Question Number : 72 Question Type : MCQ Option Shuffling : No Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

Correct Marks : 3 Wrong Marks : 0

For a perfectly mixed CSTR of volume 6 m<sup>3</sup> operating with steady state (liquid) feed rate of 0.4m<sup>3</sup>/min, what fraction of exit stream of age less than 10 min is,

- [a] 0.564      [b] 0.487      [c] 0.923      [d] 0.645

Options :

- 1. 1

2. 2

3. 3

4. 4