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Exam Date: 07-Oct-2020		
Exam lime: 15:00-18:00		
Examination: 1. Course Code - M. lech.;M.P.H.;P.G. Dinlome in Rigdete		
2. Field of Study - Nanoscience (NNST)		
SECTION 1 - Nano Science NNST		
Question No.1 (Question Id - 3) Identify the incorrect option about the wave function : (A) Continuous (B) Infinite (Correct Answer) (C) Quantized (D) Finite		
Question No.2 (Question Id - 6)		
RCONH, A RNH.		
Identify A		
Identity A		
(A) ⊖ Br ₂ /NaCl		
(B) O Br ₂ /NaOH (Correct Answer)		
$(C) \cap I_0/NaCl$		
$(D) \bigcirc I_2 (N) OH$		
Question No.3 (Question Id - 29) Ability to eat away pathogen is feature of : (A) Red blood cell (B) White blood cell (Correct Answer) (C) Liver (D) Plasma		
Question No.4 (Question Id - 41) Aceton in vapor phase is decomposed by light having a wavelength 320 nm	n yield products is :	
$(A) \bigcirc CO_2 + CH_4$		
(B) ○ CO + ·CH ₃ (Correct Answer)		
(C) \bigcirc C + CH ₄		
(D) 🔿 CO + CH ₄		
Question No.5 (Question Id - 65) Which among the following is correct about Viruses ? (A) They have DNA only (B) They have RNA only (C) They have both DNA & RNA (D) They have either DNA or RNA (Correct Answer)		
Question No.6 (Question Id - 11) When the separation between two charges is increased, the electric potenti	al energy of the charges :	
$(\Delta) \cap decreases$		
$(R) \cap \text{increases}$		
$(C) \bigcirc \text{ remains the same}$		
$(\bigcirc) \bigcirc$ remains the same		
(U) U may increase or decrease (Correct Answer)		
Question No.7 (Question Id - 38)		

In a reaction between A and B, the initial rate of reaction (r_o) was measured for different initial concentrations of A and B as given below :

A (mol L ⁻¹)	0.20	0.20	0.40
B (mol L ⁻¹)	0.30	0.10	0.05
r ₀ (mol L ⁻¹ s ⁻¹)	5.07 x 10 ⁻⁵	5.07 x 10 ⁻⁵	1.014 x 10 ⁻⁴

The order of reaction with respect to A and B are :

(A) \bigcirc A = 0, B = 1.5 (B) \bigcirc A = 1, B = 0. (Correct Answer)
(C) \bigcirc A = 2, B = 1
(D) \bigcirc A = 1, B = 1
Question No.8 (Question Id - 16) Resistance of a material in superconducting state becomes :
(A) O finite
(B) O zero (Correct Answer)
(C) O infinite
(D) 🔿 arbitrary
Question No. 9 (Question Id. 20)
The smallest unit of inheritance is called :
(A) O gene (Correct Answer)
(B) 🔿 allele
(C) 🔿 chromosome
(D) \bigcirc RNA
Question No.10 (Question Id - 48) A point charge 1 is a distance 5 cm directly above the centre of square of side 10 cm. What is the magnitude of the electric flux through the square. (ϵ_0 = permittivity of free space).
(A) O a/25
(B) $\bigcirc q/\varepsilon_0$
$(C) \bigcirc \alpha/(5 \epsilon_0)$
(D) \bigcirc q/(6 ϵ_0) (Correct Answer)
Question No.11 (Question Id - 58)
$(A) \bigcirc a sugar + a phosphate$
(A) \bigcirc a sugar ' a phosphate (B) \bigcirc a base + a sugar (Correct Answer)
(C) \bigcirc a base + a phosphate
(D) \bigcirc a base + a sugar + phosphate
Question No.12 (Question Id - 26) Name the cells which lost their control of the regulated division, differentiation and apoptosis.
(A) O Tumor cell (Correct Answer)
(B) 🔿 Immune cell
(C) O Platelets
(D) ○ Stem cells
Question No.13 (Question Id - 39)
Identify the intermediate in Reimer-Tiemann reaction :
$(A) \bigcirc - O^- Na^+ -$
CHCl ₂
(Correct Answer)
O^-Na^+
CHOCI
O^-Na^+
CCIa

Question No.14 (Question Id - 56) Thalassemia is : (A) autosome linked recessive blood disease (Correct Answer) (B) inherited metabolism disease (C) sex-linked recessive disease due to defect in red cone of eye (D) none of these
Question No.15 (Question Id - 46) X-rays of wavelength λ = a are reflected from the (100) plane of a simple cubic lattice. If the lattice constant is a, the corresponding Bragg angle (in radian) is :
 (A) ○ π/4 (B) ○ π/3 (C) ○ π/5 (D) ○ π/6 (Correct Answer)
Question No.16 (Question Id - 59) Which of the following antibody gives a primary immune reaction ?
 (A) ○ IgG (B) ○ IgM (Correct Answer) (C) ○ IgA (D) ○ IgE
Question No.17 (Question Id - 20) Following phenomenon can be explained only by particle nature of light : (A) Reflection (B) Refraction (C) Interference (D) Compton scattering (Correct Answer)
Question No.18 (Question Id - 28) What is true about genetic material of a prokaryotic cell ? (A) lacks histones (B) not enveloped by nuclear membrane (C) composed of a single circular DNA molecule (D) all of these (Correct Answer)
Question No.19 (Question Id - 1)Laplacian operator is : $(A) \bigcirc \partial/\partial x^2 + \partial/\partial y^2 + \partial/\partial z^2$ $(B) \bigcirc \partial^2/\partial xy + \partial^2/\partial yz + \partial^2/\partial xz$ $(C) \bigcirc \partial^2/\partial x^2 + \partial^2/\partial y^2 + \partial^2/\partial z^2$ (Correct Answer) $(D) \bigcirc \partial^2/\partial x + \partial^2/\partial y + \partial^2/\partial z$
Question No.20 (Question Id - 5)
This reaction is an example of : $\begin{array}{c} R \\ Na, NH_3(liq.) \\ \hline Alcohol \end{array}$ R
 (A) O Birch Reaction (Correct Answer) (B) Cannizzaro Reaction (C) Bouveault Reduction (D) Clemmensen Reduction
Question No.21 (Question Id - 9) All of the following is true statements concerning catalysts except : (A) O A catalyst will speed the rate-determining step (B) O A catalyst will be used up in a reaction (Correct Answer)

(C) \bigcirc A catalyst may induce steric strain in (D) \bigcirc A catalyst will lower the activation en	a molecule to make it re ergy of a reaction	act more readily
Question No.22 (Question Id - 15) In the superconducting state, the material beat (A) ○ ferromagnetic (B) ○ perfect diamagnetic (Correct Answer) (C) ○ ferrimagnetic (D) ○ strong paramagnetic	comes : ver)	
Question No.23 (Question Id - 21)Non-clotting of blood is caused by deficiency(A) Vitamin A(B) Vitamin C(C) Vitamin E(D) Vitamin K (Correct Answer)	of :	
Question No.24 (Question Id - 53)		
Match List - I with List - II :		
List - I	List - II	
A. Laws of motion	I. Galileo	
B. Unification of electricity and magnetism	II. Maxwell	
C. Unification of space and time	III. Newton	
D. Law of inertia	IV. Einstein	
Question No.25 (Question Id - 63) Bacteria which have ability to convert milk sur (A) Lactobacillus (Correct Answer) (B) Streptococcus (C) E. coli (D) Salmonella Question No.26 (Question Id - 27) Name the enzyme which catalyzes the oxidat (A) Transaminase (B) Glutamine synthetase	gar in to lactic acid is ca	led :
(C) O Phosphofructokinase		
(D) Oxidoreductase (Correct Answer)		
Question No.27 (Question Id - 24)Which ratio is constant for DNA ?(A) \bigcirc A + G/T + C (Correct Answer)(B) \bigcirc A + T/G + C(C) \bigcirc A + C/U + G(D) \bigcirc A + U/G + C		
Question No.28 (Question Id - 32) For the oxidation of iron 4Fe(s) + 3O ₂ (g) → 2Fe ₂ O ₃ (s) Entropy change is - 549.4 J K ⁻¹ mol ⁻¹ at 298 I (Δ _r H ^Θ = - 1648 x 10 ³ J mol ⁻¹) (A) ○ 4.98 kJ K ⁻¹ mol ⁻¹ (Correct Answer) (B) ○ 5.53 kJ K ⁻¹ mol ⁻¹ (C) ○ 6.07 kJ K ⁻¹ mol ⁻¹ (D) ○ 5.23 kJ K ⁻¹ mol ⁻¹	K. Total entropy change	for this reaction is :
Question No.29 (Question Id - 43) Let P and E denote the linear momentum and	d energy of a photon. If t	ne wavelength is decreased,

(A) $\bigcirc\,$ Both P and E increase (Correct Answer)

(B) $\bigcirc\,$ P increases and E decreases

(C) \bigcirc P decreases and E increases

Question No.30 (Question Id - 31)

 $^{238}_{92}$ Uundergo alpha (α) decay form : (A) $\bigcirc \frac{234}{90}$ Th $+ \frac{4}{2}\alpha$ (Correct Answer) (B) $\bigcirc \frac{236}{90} \text{Th} + \frac{2}{2} \alpha$ (C) $\bigcirc \frac{234}{88} Th + \frac{4}{4} \alpha$ (D) $\bigcirc \frac{236}{88} Th + \frac{2}{4} \alpha$ Question No.31 (Question Id - 17) With increase in temperature, the orientation polarization in general : (A) 🔘 increases (B) O decreases (Correct Answer) (C) 🔘 is constant (D) O none of these Question No.32 (Question Id - 44) Width of one-dimensional infinite potential well is decreased by half, what is the effect on its energy levels. (A) O Energy levels do not change (B) O Energy of the levels is doubled (C) O Energy of the levels is quadrupled (Correct Answer) (D) O Energy of the levels is halved Question No.33 (Question Id - 25) Which of the following promotes hypoglycemia ? (A) O Epinephrine (B) O Insulin (Correct Answer) (C) O Norepinephrine (D) O Growth hormone Question No.34 (Question Id - 33) In tetrathionate ion (S₄O₆²⁻), the oxidation numbers of sulphurs is/are : (A) O, + 5 (Correct Answer) (B) 🔘 2.5 (C) 🔿 + 2, + 3 (D) 🔿 + 3 Question No.35 (Question Id - 64) Law of thermodynamics which states that energy can neither be created nor be destroyed is (A) \bigcirc The second law of thermodynamics $(B) \bigcirc$ Third law of thermodynamics (C) O First law of thermodynamics (Correct Answer) (D) O Zero-order kinetics Question No.36 (Question Id - 23) Programmed cell death can be termed as : (A) Oxidative stress (B) O Apoptosis (Correct Answer) (C) O Cell division (D) 🔿 Cell cycle Question No.37 (Question Id - 50) In the polarization vs field strength for a ferroelectric material, Ps stands for : (A) O Space charge polarization (B) O Spontaneous polarization (Correct Answer) (C) O Saturation polarization (D) O None of these Question No.38 (Question Id - 54)

(A) ○ ATGCATCACA (A) ○ ATGCATCACA (C) ○ TAGCTACGT (Correct Answer) (D) ○ Debye temperature Coestion No.41 (Coestion 14 - 2) Which of the following molecule is tigonal pyramid shape ? (A) ○ †b(○ (D) ○ DFG Coestion No.41 (Coestion 14 - 19) There is any protein to an ideal gas in an isothermal process. (A) ○ †b or said process is not possible Coestion No.42 (Coestion 14 - 30) The easier and protein to consistive work. (D) ○ the gas will do positive work. (C) ○ the gas will do positive work. (C) ○ The gas will do positive work. (C) ○ The gas will do positive work. (D) ○ the gas will do positive work.	complementary strand ?
 (A) ⊂ ATGCATGCA (B) ⊂ AUGCAGGCA (C) ⊂ TAGGTAGGT (Greet Answer) (D) ∪ UACGUACGU Covertion No.32 (Question 14 - 12) The temperature of antiferromagnetic to paramagnetic transition is called : (A) ⊂ Antiferromagnetic Curle temperature (B) ⊂ Curre-West temperature (C) ⊂ Neel temperature (N) ⊂ Net (Correct Answer) (C) ⊂ D temperature derive oth (D) ⊂ the said process is not possible Duestion No.42 (Question 14 - 31 The metar solubility of N(Christ in 0.1 M NaOH is (onic product of N(CH) ₂ = 2.0 x 10 ⁻⁴⁵): (A) ⊂ 1 A 10⁻¹⁵ M (B) ⊂ 2 0 x 10⁻¹³ M Duestion No.42 (Question 14 - 37 Wein below are two statements: one is labeled as Assertion A and the other is labeled as Reason R: Assertion A: Naserion 1: Neit Correct Answer) (C) ⊂ A is ordered that 15 MO The correct captions of A (B) ⊂ Mo Correct Answer) (C) △ A is not correct and R is the correct explanation of A (Correct Answer) (C) △ A is not correct that R is ordered (N) ⊂ A is not correct answer) (C) △ A i	
(B) ⊂ AUGCAUGCA (C) TACGUACGU (Correct Answer) (D) ⊂ UACGUACGU Outsion No.33 (Question 1d - 13) The temperature of antiferromagnetic fourie temperature (B) ⊂ Cub-Works Integrature (C) ⊂ Neal temperature (Correct Answer) (D) ⊂ Deby temperature Coursion No.40 (Question 1d - 2) Which of the following molecule is trigonal prysmid shape ? (A) ⊂ H ₂ () (D) ⊂ Deby temperature Coursion No.40 (Question 1d - 2) Which of the following molecule is trigonal prysmid shape ? (A) ⊂ H ₂ () (D) ⊂ Deby temperature Coursion No.40 (Question 1d - 19) If heat is supplied to an ideal gas in an isofhermal process, (A) ⊂ H ₂ () (D) ⊂ DBy temperature Coursion No.41 (Question 1d - 19) If heat is supplied to an ideal gas in an isofhermal process, (A) ⊂ H ₂ (C) ← MH ₃ (Correct Answer) (D) ⊂ BF ₃ Cuestion No.42 (Question 1d - 19) If heat is supplied to an ideal gas in an isofhermal process, (A) ⊂ H ₂ (C) ← MH ₃ (C) ⊂ C) (D) ⊂ DBy temperature Cuestion No.42 (Question 1d - 19) Cuestion No.42 (Question 1d - 34) Cuestion No.42 (Question 1d - 34) Cuestion No.42 (Question 1d - 37) Civen below are two statements : one is labeled as Assertion A and the other is labeled as Reason R : Assertion A: Reason R: (A) ⊂ Dist A correct Answer) (C) ⊂ A ki corr ¹⁵ M Cuestion No.43 (Question 1d - 77) Civen below are two statements : one is labeled as Assertion A and the other is labeled as Reason R : Assertion A: Reason R: (A) ⊂ Dist A and R are correct and R is the correct osplanation of A (Correct Answer) (C) ⊂ A is not correct I and R is the correct explanation of A (Correct Answer) (C) ⊂ A is not correct I at R is MC the ocrect osplanation of A (Correct Answer) (C) ⊂ A is not correct I at R is MC the ocrect osplanation of A (Correct Answer) (C) ⊂ A is not correct I at R is MC the correct osplanation of A (Correct Answer) (C) ⊂ A is not correct I at R is MC the correct osplanation of A (Correct Answer) (C) ⊂ A is ordered I at R is ordered I as Meant I asesentially c	(A) 🔿 ATGCATGCA
 (C) □ TACGTACGT (Correct Answer) (D) □ UACGUACGU Ouestion No.39 (Question Id - 13) The temperature of antiferromagnetic transition is called : (A) □ Antiferromagnetic Curie temperature (B) □ Curie-Weiss temperature (C) Neel temperature (Correct Answer) (D) □ Debye temperature Question No.49 (Question Id - 2) Which of the following molecule is tingonal pyramid shape ? (A) □ Antiferromagnetic Curie tansver) (D) □ Debye temperature Question No.41 (Question Id - 19) If heat is supplied to an local gas in an isothermal process, (A) □ the internal energy of the gas will decrease (B) □ to gas will do positive work (Correct Answer) (C) □ the gas will do positive work (Correct Answer) (C) □ the gas will do positive work (Correct Answer) (C) □ the gas will do positive work (Correct Answer) (C) □ the gas will do positive work (Correct Answer) (C) □ the gas will do positive work (Correct Answer) (C) □ the gas will do positive work (Correct Answer) (C) □ the gas will do positive work (Correct Answer) (C) □ the gas will do positive work (Correct Answer) (C) □ the gas will do positive work (Correct Answer) (C) □ the gas will do positive work (Correct Answer) (C) □ the gas will do positive work (Correct Answer) (C) □ the gas will do positive work (Correct Answer) (C) □ the gas will do positive work (Correct Answer) (C) □ the gas will conset to the substeaments : one is labeled as Assertion A and the other is labeled as Reason R : Assertion A: Reason R: In the representation E⁴(Fe²⁺¹, Fe²⁺¹) and E⁶(Cu²⁺¹Cu), (Fe³⁺Fe²⁺¹) and Cu²⁺¹Cu) are redox couple. In the light of the above statements : choose the most appropriate answer from the coptions given balow : (A) □ Goti A and R are correct explanation of A	(B) O AUGCAUGCA
(1) © UACSUNCESU Question No.39 (Question 1d - 13) The temporature of antiferromagnetic transition is called : (A) ○ Antiferromagnetic Curle temporature (B) ○ Curle-Weiss temporature (C) Net temporature (Correct Answer) (D) ○ Debye temperature Question No.40 (Question 1d - 2) Which of the following metacula is trigonal pyramid shape ? (A) ○ H_0 (B) ○ CQ2 (C) NH3 (Correct Answer) (D) ○ DF3 Question No.41 (Question 1d - 19) Pheat is supplied boan riskal gas in a insthemal process, (A) ○ the internal energy of the gas will decrease (B) ○ the gas will do positive work (Correct Answer) (C) ○ the gas will do positive work (Correct Answer) (C) ○ the gas will do positive work (Correct Answer) (C) ○ the said process is not possible Question No.42 (Question 1d - 19) Pheat is supplied boan riskal gas in an isothermal process, (A) ○ the internal energy of the gas will docrease (B) ○ the gas will do positive work (Correct Answer) (C) ○ the gas will do positive work (Correct Answer) (C) ○ the gas will do positive work (Correct Answer) (C) ○ 50 × 10 ¹² M (D) ○ 40 × 10 ¹² M Question No.42 (Question 1d - 37) Course level was tataments: one is tabelled as Assertion A and the other is tabelled as Reason R : Assertion A: Readox couple is the combination of oxid/zed and reduced form of a substance involved in an oxidation or eduction natione. Reason R: In the representation E ² (Fe ²⁺ , Fe ²⁺) and E ⁰ (Cu ²⁺ (Cu), (Fe ³⁺ , Fe ²⁺) and (Cu ²⁺ /Cu) are redox couple. In the representation E ² (Fe ²⁺ , Fe ²⁺) and E ⁰ (Cu ²⁺ (Cu), (Fe ³⁺ , Fe ²⁺) and (Cu ²⁺ /Cu) are redox couple. In the representation E ³ (Fe ²⁺) and E ⁶ (Cu ²⁺ /Cu), (Fe ³⁺ , Fe ²⁺) and (Cu ²⁺ /Cu) are redox couple. In the representation E ³ (Fe ²⁺) and E ⁶ (Cu ²⁺ /Cu), (Fe ³⁺ , Fe ²⁺) and (Cu ²⁺ /Cu) are redox couple. In the representation E ³ (Fe ²⁺) and E ⁶ (Cu ²⁺ /Cu), (Fe ³⁺ , Fe ²⁺) and (Cu ²⁺ /Cu) are redox couple. In the representation E ³ (Fe ²⁺) and E ⁶ (Cu ²⁺ /Cu), (Fe ³⁺ , Fe ²⁺) and (Cu ²⁺ /Cu) are red	(C) O TACGTACGT (Correct Answer)
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(A) △ Antiferromagnetic Curie temperature (A) △ Antiferromagnetic Curie temperature (C) □ Antiferromagnetic Curie temperature (C) □ Neet temperature (C) □ Neet temperature (C) □ Detye temperature (D) □ Detyee temperature	Question No.39 (Question Id - 13)
 (A) ○ Antiferromagnetic Curle temperature (B) ○ Curle Vess temperature (C) ○ Neel temperature (C) ○ Neel temperature (C) ○ Neel temperature (C) ○ Debye temperature (C) ○ Neel temperature (C) ○ Neel temperature (C) ○ Neel temperature (C) ○ Neth (Correct Answer) (D) ○ EF₃ Cureation No.41 (Question Id - 19) (Theat is supplied to an ideal gas in an isothermal process. (A) ○ the internal energy of the gas will decrease (B) ○ the gas will do positive work (Correct Answer) (C) ○ the gas will do positive work (Correct Answer) (C) ○ the gas will do negative work (D) ○ the said process is not possible Cuestion No.42 (Question Id - 34) The moder solubility of NICH1; in 0.1 M NAOH is (ionic product of Ni(OH) ₂ = 2.0 x 10 ⁻¹⁶): (A) ○ 10 x 10⁻¹² M (B) ○ 20 x 10⁻¹³ M Cuestion No.43 (Question Id - 37) Given below are two statements: one is labelled as Assertion A and the other is labelled as Reason R : Assertion A: Reason R: In the representation E ⁰ (Fe ³⁺ /Fe ²⁺) and E ⁰ (Cu ²⁺ /Cu), (Fe ³⁺ /Fe ²⁺) and (Cu ²⁺ /Cu) are redox couple. In the light of the above statements. choose the most appropriate answer from the options given below : (A) ○ Doth A and R are correct and R is the correct explanation of A (Correct Answer) (C) ○ 4 is correct but R is NOT the correct explanation of A (Correct Answer) (C) ○ del wall Cuestion No.45 (Question Id - 5) Innor membrane of which organelle forms infolding called cristae ? (A) ○ is borned to use a labelled form is indeding called cristae ? (A) ○ cell wall Cuestion No.45 (Question Id - 6) Identify the oxide of anlance ant metal which is essentially covalent in nature : (A) ○ cell wall Cuestion No.45 (Question Id - 6) Identify the oxi	The temperature of antinerromagnetic to paramagnetic transition is called .
 (6) Curie-Weiss temperature (Correct Answer) (7) Debye temperature Ouestion No.49 (Question 14 - 2) Which of this following molecule is trigonal pyramid shape ? (A) D H₂O (B) C C2 (C) N H4 (Correct Answer) (D) BF3 Cuestion No.41 (Question 14 - 19) If heat is supplied to an ideal gas in an isothermal process, (A) T the internal energy of the gas will decrease (B) C the gas will do positive work (Correct Answer) (C) the gas will do positive work (Correct Answer) (C) the task will do negative work (D) the said process is not possible Cuestion No.42 (Question 14 - 34) The moler solubility of NI(Oth) in 0.1 M NaOH Is (ionic product of Ni(OH)₂ = 2.0 x 10⁻¹⁵): (A) 1 for 10⁻² M (B) 2 0 x 10⁻¹² M (C) to the gas will do creative work (C) 5 0 x 10⁻¹² M (D) 4 0 x 10⁻¹³ M (D) 4 0 x 10⁻¹³ M (Correct Answer) (C) 5 0 x 10⁻¹² M (D) 4 0 x 10⁻¹³ M (D) 4 0 x 10	(A) 🔿 Antiferromagnetic Curie temperature
(□) □ Neel temperature (Correct Answer) (□) □ Debye temperature Cuestion No.40 (Question 1d - 2) Which of the following molecule is trigonal pyramid shape ? (A) □ H ₂ O (□) □ Debye temperature (□) □ Debye temperature (□) □ DF3 Cuestion No.41 (Question 1d - 19) If heat is supplied to an ideal gas in an isothermal process. (A) □ the internal hearg of the gas will do positive work (Correct Answer) (□) □ the gas will do positive work (Correct Answer) (□) □ the gas will do positive work (Correct Answer) (□) □ the gas will do positive work (Correct Answer) (□) □ the gas will do positive work (Correct Answer) (□) □ the gas will do positive work (Correct Answer) (□) □ the gas will do positive work (Correct Answer) (□) □ the gas will do positive work (Correct Answer) (□) □ the gas will do positive work (Correct Answer) (□) □ the gas will do positive work (Correct Answer) (□) □ the gas will do positive work (Correct Answer) (□) □ the the bothermal process in tho do positite work (Correct Answer)	(B) Curie-Weiss temperature
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Question No.49 (Question Id - 2) Which of the following molecule is trigonal pyramid shape ? (A) H ₂ O (B) CO2_ (C) NH ₃ (Correct Answer) (D) BF3 Question No.41 (Question Id - 19) If heat is supplied to an ideal gas in an isothermal process, (A) the intermal energy of the gas will decrease (B) the gas will do positive work (Correct Answer) (C) the gas will do positive work (D) the said process is not possible Question No.42 (Question Id - 34) The molar solubility of NI(OH) ₂ in 0.1 M NaOH is (ionic product of NI(OH) ₂ = 2.0 x 10 ⁻¹⁵): (A) 1 to x 10 ⁻¹² M (B) 2.0 x 10 ⁻¹³ M (Correct Answer) (C) 5.0 x 10 ⁻¹² M (B) 2.0 x 10 ⁻¹³ M (Correct Answer) (C) 5.0 x 10 ⁻¹² M (D) 4.0 x 10 ⁻¹³ M 21/Cu) are redox couple.	
The transformation of the control indecade is ingenial product of the control of the control is a set of the control of control of the control of the control of the control of the contr	Question No.40 (Question Id - 2) Which of the following molecule is trigonal ovramid shape 2
 (B) ⊂ CO₂ (C) → NH₃ (Correct Answer) (D) ⇒ BF₃ Cuestion No.41 (Question Id - 19) If heat is supplied to an ideal gais in an isothermal process, (A) → the internal energy of the gas will decrease (B) → the gas will do negative work (Correct Answer) (C) → the gas will do negative work (Correct Answer) (C) → the said process is not possible Cuestion No.42 (Question Id - 34) The molar solubility of Ni(OH)₂ in 0.1 M NaOH is (ionic product of Ni(OH)₂ = 2.0 x 10⁻¹⁵): (A) → 1.0 x 10⁻¹² M (B) → 2.0 x 10⁻¹³ M (C) → 5.0 x 10⁻¹³ M (D) → 4.0 x 10⁻¹³ M (C) → 5.0 x 10⁻¹⁴ M (D) → 4.0 x 10⁻¹³ M (Exaction No.43 (Question Id - 37) (Given below are two statements : one is labelled as Assertion A and the other is labelled as Reason R : Assertion A: Reason R: In the representation E^(D)(Fe³⁺, Fe²⁺) and E^(D)(Cu²⁺/Cu), (Fe³⁺/Fe²⁺) and (Cu²⁺/Cu) are redox couple. In the light of the above statements, choose the most appropriate answer from the options given below : (A) ○ Both A and R are correct and R is the correct explanation of A (Correct Answer) (C) ○ A is not correct but R is not correct (A) ○ Is not correct It R is not correct (B) ○ mitochordinal (Correct Answer) (C) ○ golgi apparatus (D) ○ call wall Cuestion No.45 (Question Id - 9) Identify the oxide of alkaline east the metal which is esse	(A) \bigcirc H ₂ O
<pre>(C) ○ NH₃ (Correct Answer) (D) ○ BF₃ Question Nc.41 (Question Id - 19) If here its supplied to an ideal pais in an isothermal process. (A) ○ the gas will do positive work (Correct Answer) (C) ○ the gas will do positive work (Correct Answer) (C) ○ the gas will do positive work (Correct Answer) (C) ○ the said process is not possible Question No.42 (Question Id - 34) The molar solubility of N(OH); in 0.1 N NaOH is (ionic product of Ni(OH); = 2.0 x 10⁻¹⁵): (A) ○ 1.0 × 10⁻¹² M (B) ○ 2.0 x 10⁻¹³ M (Correct Answer) (C) ○ 5.0 x 10⁻¹² M (D) ○ 4.0 x 10⁻¹³ M (C) ○ 4.0 x 10⁻¹⁴ M (C) ○ 4.0 x 10⁻¹⁴ M (C) ○ 4.0 x 10⁻¹⁵ M (C) ○ 4.0 x 10⁻¹⁵ M (C) ○ 4.0 x 10⁻¹⁵ M (C) ○ 4.0 x 10⁻¹⁶ M (C) ○ 4.1 s correct but R is the correct explanation of A (C) ○ 6.0 th A and R are correct and R is the correct explanation of A (Correct Answer) (C) ○ A is not correct but R is NOT the correct explanation of A (Correct Answer) (C) ○ A is not correct but R is not correct (D) ○ A is not correct but R is not correct (D) ○ A is not correct but R is not correct (C) ○ A is correct but R is not correct (C) ○ A is not correct but R is not correct (C) ○ A is not correct but R is not correct (C) ○ A is not correct but R is not correct (C) ○ A is not correct but R is not correct (C) ○ A is not correct but R is not correct (C) ○ A is not correct but R is not correct (C) ○ A is not correct but R is not correct (C) ○ A is not correct but R is not correct (C) ○ A is not correct but R is not correct (C) ○ A is not correct but R is not correct (D) ○ the wall Cuestion No.45 (Question Id - 5) Iterify the outpoint of correct Answer) (C) ○ cell wall Cuestion No.45 (Question Id - 6) (D) ○ SrO</pre>	$(B) \bigcirc CO_2$
 (D) □ BF₃ Cuestion No.41 (Question Id - 19) If the stat is supplied to an ideal gas in an isothermal process, (A) □ the itemanal energy of the gas will do positive work (Correct Answer) (C) □ the gas will do nogative work (Correct Answer) (C) □ the said process is not possible Cuestion No.42 (Question Id - 34) The molar solubility of Ni(OH)₂ in 0.1 M NaOH is (ionic product of Ni(OH)₂ = 2.0 x 10⁻¹⁵): (A) □ the internal energy of Ni Correct Answer) (C) □ the said process is not possible Cuestion No.42 (Question Id - 34) The molar solubility of Ni(OH)₂ in 0.1 M NaOH is (ionic product of Ni(OH)₂ = 2.0 x 10⁻¹⁵): (A) □ 1.0 x 10⁻¹² M (B) □ 2.0 x 10⁻¹³ M Cuestion No.43 (Question Id - 37) Given below are two statements: one is labelled as Assertion A and the other is labelled as Reason R: Assertion A: Redux couple is the combination of oxidized and reduced form of a substance involved in an oxidation or reduction half cell. Reason R: In the representation E⁰(<i>Fe</i>²⁺ /<i>Fe</i>²⁺) and E⁰ (Cu²⁺/Cu), (<i>Fe</i>²⁺ /<i>Fe</i>²⁺) and (Cu²⁺/Cu) are redox couple. In the light of the above statements, choose the most appropriate answer from the options given below : (A) □ Both A and R are correct but R is NOT the correct explanation of A (Correct Answer) (C) □ A is not correct (B) □ mictochondria (Correct Answer) (C) □ do and R are correct but R is NOT the correct explanation of A (Correct Answer) (C) □ goig apparatus (B) □ mictochondria (Correct Answer) (C) □ goig apparatus (P) □ Correct Answer) (C) □ goig apparatus (P) □ Correct Answer) (C) □ goig apparatus (P) □ Goid (Correct Answer) (C) □ goig apparatus (P) □ Goid (P) □ SrO	(C) O NH ₃ (Correct Answer)
Question No.41 (Question Id - 19) If heat is supplied to an ideal gas in an isothermal process. (A) the internal energy of the gas will do crease (B) the gas will do negative work (Correct Answer) (C) the said process is not possible Question No.42 (Question Id - 34) The molar solubility of Ni(OH) ₂ in 0.1 M NaOH is (ionic product of Ni(OH) ₂ = 2.0 x 10 ⁻¹⁵): (A) 1.0 x 10 ⁻¹² M (B) 5.0 x 10 ⁻¹³ M Question No.43 (Question Id - 37) Given below are two statements: one is labelled as Assertion A and the other is labelled as Reason R : Assertion A: Reason R: In the representation E ⁰ (Fe ³⁺ /Fe ²⁺) and E ⁰ (Cu ²⁺ /Cu), (Fe ³⁺ /Fe ²⁺) and (Cu ²⁺ /Cu) are redox couple. In the representation E ⁰ (Fe ³⁺ /Fe ²⁺) and E ⁰ (Cu ²⁺ /Cu), (Fe ³⁺ /Fe ²⁺) and (Cu ²⁺ /Cu) are redox couple. In the light of the above statements, choose the most appropriate answer from the options given below : (A) Both A and R are correct but R is NOT the correct explanation of A (Correct Answer) (C) \triangle A is not correct (C) (D) \triangle A is not correct but R is not correct (D) \triangle A is not correct but R is not correct (D) \triangle A is not correct but R is not correct explanation of A (Correct Answer) ((D) O BF ₃
Healt is updated. Inc. I: (protect in the original process). (A) the internal energy of the gas will decrease (A) the internal energy of the gas will decrease (B) the gas will do negative work (D) the said process is not possible Question No.42 (Question Id - 34) The molar solubility of NI(OH) ₂ in 0.1 M NaOH is (ionic product of Ni(OH) ₂ = 2.0 x 10 ⁻¹⁶) : (A) 1 ox 10 ⁻¹² M (B) 2.0 x 10 ⁻¹³ M (Correct Answer) (C) 5.0 x 10 ⁻¹² M Question No.43 (Question Id - 37) Given below are two statements : one is labelled as Assertion A and the other is labelled as Reason R : Assertion A: Redox couple is the combination of oxidized and reduced form of a substance involved in an oxidation or reduction half cell. Reason R: In the representation E [®] (Fe ³⁺ /Fe ²⁺) and E [®] (Cu ²⁺ /Cu), (Fe ³⁺ /Fe ²⁺) and (Cu ²⁺ /Cu) are redox couple. In the light of the above statements, choose the most appropriate answer from the options given below : (A) Both A and R are correct and R is the correct explanation of A (Correct Answer) (C) A is not correct. (A) Both A and R are correct and R is the correct explanation of A (Correct Answer) (C) A is not correct. <td>Question No. 41 (Question Id - 19)</td>	Question No. 41 (Question Id - 19)
 (A) □ the internal energy of the gas will decrease (B) □ the gas will do positive work (Correct Answer) (C) □ the gas will do positive work (Correct Answer) (C) □ the said process is not possible Question No.42 (Question Id - 34) The molar solubility of Ni(OH) ₂ in 0.1 M NaOH is (ionic product of Ni(OH) ₂ = 2.0 x 10 ⁻¹⁵): (A) □ 1.0 x 10⁻¹² M (B) □ 2.0 x 10⁻¹³ M Question No.43 (Question Id - 37) Given below are two statements: one is labelled as Assertion A and the other is labelled as Reason R : Assertion A: Redox couple is the combination of oxidized and reduced form of a substance involved in an oxidation or reduction half cell. Reason R: In the representation E ^O (Fe ³⁺ /Fe ²⁺) and E ^O (Cu ²⁺ /Cu), (Fe ³⁺ /Fe ²⁺) and (Cu ²⁺ /Cu) are redox couple. In the light of the above statements, choose the most appropriate answer from the options given below : (A) © Both A and R are correct and R is the correct explanation of A (B) © Both A and R are correct but R is NOT the correct explanation of A (Correct Answer) (C) △ A is nort correct but R is not correct Question No.44 (Question Id - 55) Inner membrane of which organelle forms infolding called cristae ? (A) □ toksomes (B) □ mitochondria (Correct Answer) (C) ○ cell wall Question No.45 (Question Id - 8) Identify the oxide of alkaline earth metal which is essentially covalent in nature : (A) □ BoQ (Correct Answer) (B) □ Sr0 (C) ○ Sr0 (C) ○ Sr0 (D) ○ Sr0 (D) ○ Sr0	If heat is supplied to an ideal gas in an isothermal process,
 (B) C the gas will do positive work (Correct Answer) (C) the gas will do positive work (D) C the said process is not possible Question No.42 (Question Id - 34) The molar solubility of Ni(OH)₂ in 0.1 M NaOH is (ionic product of Ni(OH)₂ = 2.0 x 10⁻¹⁵): (A) 1.0 x 10⁻¹² M (B) 2.0 x 10⁻¹³ M (D) 4.0 x 10⁻¹³ M Question No.43 (Question Id - 37) Given below are two statements : one is labelled as Assertion A and the other is labelled as Reason R : Assertion A: Redox couple is the combination of oxidized and reduced form of a substance involved in an oxidation or reduction half cell. Reason R: In the representation E⁹(Fe³⁺ /Fe²⁺) and E⁹ (Cu²⁺/Cu), (Fe³⁺/Fe²⁺) and (Cu²⁺/Cu) are redox couple. In the light of the above statements, choose the most appropriate answer from the options given below : (A) Both A and R are correct and R is the correct explanation of A (Correct Answer) (C) A is not correct (D) A is not correct but R is NOT the correct explanation of A (Correct Answer) (C) A is not correct but R is correct Question No.44 (Question Id - 55) Inner membrane of which organelle forms infolding called cristae ? (A) mitchondria (Correct Answer) (C) O is correct but R is correct Question No.45 (Question Id - 8) Identify the code of alkaline earth metal which is essentially covalent in nature : (A) Be0 (Correct Answer) (B) Mg0 (C) C ao (D) C sr0 	(A) O the internal energy of the gas will decrease
 (C) be gas will do negative work (D) the said process is not possible Question No.42 (Question Id - 34) The molar solubility of NI(OH)₂ in 0.1 M NaOH is (ionic product of Ni(OH)₂ = 2.0 x 10⁻¹⁵): (A) 1.0 x 10⁻¹² M (B) 2.0 x 10⁻¹³ M (Correct Answer) (C) 5.0 x 10⁻¹² M (D) 4.0 x 10⁻¹³ M Question No.43 (Question Id - 37) Given below are two statements : one is labelled as Assertion A and the other is labelled as Reason R : Assertion A: Redox couple is the combination of oxidized and reduced form of a substance involved in an oxidation or reduction half cell. Reason R: In the representation E^Θ(Fe³⁺ /Fe²⁺) and E^Θ (Cu²⁺/Cu), (Fe³⁺/Fe²⁺) and (Cu²⁺/Cu) are redox couple. In the light of the above statements, choose the most appropriate answer from the options given below : (A) Both A and R are correct and R is the correct explanation of A (B) Both A and R are correct but R is NOT the correct explanation of A (Correct Answer) (C) A is correct but R is correct (D) A is not correct but R is correct (D) A is not correct but R is correct (D) A is not correct Answer) (C) go gig apparatus (D) cell wall Question No.43 (Question Id - 6) Identify the oxide of alkaline earth metal which is essentially covalent in nature : (A) BoD (Correct Answer) (B) Go (C) C aO (C) C aO 	(B) ○ the gas will do positive work (Correct Answer)
(U) → the said process is not possible Question No.42 (Question Id - 34) The molar solubility of NI(OH) ₂ in 0.1 M NaOH is (ionic product of NI(OH) ₂ = 2.0 x 10 ⁻¹⁵) : (A) → 1.0 x 10 ⁻¹² M (B) → 2.0 x 10 ⁻¹³ M Question No.43 (Question Id - 37) Given below are two statements : one is labelled as Assertion A and the other is labelled as Reason R : Assertion A: Redox couple is the combination of oxidized and reduced form of a substance involved in an oxidation or reduction half cell. Reason R: In the representation $E^{\Theta}(Fe^{3+}/Fe^{2+})$ and $E^{\Theta}(Cu^{2+}/Cu)$, (Fe^{3+}/Fe^{2+}) and (Cu^{2+}/Cu) are redox couple. In the isponse statements, choose the most appropriate answer from the options given below : (A) → Both A and R are correct and R is the correct explanation of A (Correct Answer) (C) → A is not correct but R is not correct (D) → A is not correct but R is not correct (D) → A is not correct that R is correct (D) → A is not correct Answer) (D) → cell wall Question No.45 (Question Id - 50) Intermembrane of which organelle forms infolding called cristae ? (A) → indochondria (Correct Answer) (D) → cell wall Question No.45 (Question Id - 6) Identify the oxide of alkaline earth metal which is essentially covalent in nature : (A) → BoO (Correct Answer) (B) → MgO (C) → CaO (D) → SirO	(C) ○ the gas will do negative work
Question No.42 (Question Id - 34) The molar solubility of NI(OH) ₂ in 0.1 M NaOH is (ionic product of NI(OH) ₂ = 2.0 x 10 ⁻¹⁵): (A) \bigcirc 1.0 x 10 ⁻¹² M (B) \bigcirc 2.0 x 10 ⁻¹³ M (C) \bigcirc 5.0 x 10 ⁻¹² M (D) \bigcirc 4.0 x 10 ⁻¹³ M Question No.43 (Question Id - 37) Given below are two statements : one is labelled as Assertion A and the other is labelled as Reason R : Assertion A: Redox couple is the combination of oxidized and reduced form of a substance involved in an oxidation or reduction half cell. Reason R: In the representation E ⁰ (Fe ³⁺ /Fe ²⁺) and E ⁰ (Cu ²⁺ /Cu), (Fe ³⁺ /Fe ²⁺) and (Cu ²⁺ /Cu) are redox couple. In the representation E ⁰ (Fe ³⁺ /Fe ²⁺) and E ⁰ (Cu ²⁺ /Cu), (Fe ³⁺ /Fe ²⁺) and (Cu ²⁺ /Cu) are redox couple. In the light of the above statements, choose the most appropriate answer from the options given below : (A) O Both A and R are correct and R is the correct explanation of A (B) O Both A and R are correct but R is NOT the correct explanation of A (Correct Answer) (C) O A is not correct but R is not correct (D) O A is not correct but R is correct (P) O at is not correct but R is correct (P) O cell wall Cuestion No.45 (Question Id - 6) Iternify the oxide of alkaline earth metal which is essentially covalent in nature :	(D) () the said process is not possible
The molar solubility of NI(OH) ₂ in 0.1 M NaOH is (ionic product of NI(OH) ₂ = 2.0 x 10 ⁻¹⁵) : (A) $0 = 1.0 \times 10^{-12} M$ (B) $2.0 \times 10^{-13} M$ Question No.43 (Question Id - 37) Given below are two statements : one is labelled as Assertion A and the other is labelled as Reason R : Assertion A: Redox couple is the combination of oxidized and reduced form of a substance involved in an oxidation or reduction haif cell. Reason R: In the representation E ⁰ (Fe ³⁺ /Fe ²⁺) and E ⁰ (Cu ²⁺ /Cu), (Fe ³⁺ /Fe ²⁺) and (Cu ²⁺ /Cu) are redox couple. In the light of the above statements, choose the most appropriate answer from the options given below : (A) \bigcirc Both A and R are correct and R is the correct explanation of A (B) \bigcirc Both A and R are correct but R is NOT the correct explanation of A (Correct Answer) (C) \triangle A is not correct but R is not correct (D) \triangle A is not correct but R is correct Question No.44 (Question Id - 55) Inner membrane of which organelle forms infolding called cristae ? (A) \bigcirc both Correct Answer) (C) \bigcirc golgi apparatus (D) \bigcirc cell wall Question No.45 (Question Id - 6) Identify the oxide of alkaline earth metal which is essentially covalent in nature : (A) \bigcirc Bodo (Correct Answer) (B) \bigcirc MigO (C) \bigcirc CaO (D) \bigcirc S rO	Question No.42 (Question Id - 34)
 (A) ○ 1.0 × 10⁻¹² M (B) ○ 2.0 × 10⁻¹² M (C) ○ 5.0 × 10⁻¹² M (D) ○ 4.0 × 10⁻¹³ M Question No.43 (Question Id - 37) Given below are two statements : one is labelled as Assertion A and the other is labelled as Reason R : Assertion A: Redox couple is the combination of oxidized and reduced form of a substance involved in an oxidation or reduction half cell. Reason R: In the representation E⁹(Fe³⁺ /Fe²⁺) and E⁹ (Cu²⁺/Cu), (Fe³⁺/Fe²⁺) and (Cu²⁺/Cu) are redox couple. In the light of the above statements, choose the most appropriate answer from the options given below : (A) ○ Both A and R are correct and R is the correct explanation of A (B) ○ Both A and R are correct but R is NOT the correct explanation of A (Correct Answer) (C) ○ A is correct but R is correct Question No.44 (Question Id - 55) Inner membrane of which organelle forms infolding called cristae ? (A) ○ inbosomes (B) ○ mitochondria (Correct Answer) (C) ○ golgi apparatus (D) ○ cell wall Question No.45 (Question Id - 6) Identify the oxide of alkaline earth metal which is essentially covalent in nature : (A) ○ BotO (Correct Answer) (B) ○ Mg0 (C) ○ CaO (D) ○ SrO 	The molar solubility of Ni(OH) ₂ in 0.1 M NaOH is (ionic product of Ni(OH) ₂ = 2.0 x 10^{-15}):
 (B) 2.0 × 10⁻¹³ M (Correct Answer) (C) 5.0 × 10⁻¹² M (D) 4.0 × 10⁻¹³ M Cuestion No.43 (Question Id - 37) Given below are two statements : one is labelled as Assertion A and the other is labelled as Reason R : Assertion A: Redox couple is the combination of oxidized and reduced form of a substance involved in an oxidation or reduction half cell. Reason R: In the representation E^Θ(Fe³⁺ /Fe²⁺) and E^Θ (Cu²⁺/Cu), (Fe³⁺/Fe²⁺) and (Cu²⁺/Cu) are redox couple. In the light of the above statements, choose the most appropriate answer from the options given below : (A) Both A and R are correct and R is the correct explanation of A (B) Both A and R are correct but R is NOT the correct explanation of A (Correct Answer) (C) A is correct but R is not correct (D) A is not correct but R is correct Question No.44 (Question Id - 55) Inner membrane of which organelle forms infolding called cristae ? (A) or ibosomes (B) mitochondria (Correct Answer) (C) oglogi apparatus (D) cell wall Question No.45 (Question Id - 8) Identify the oxide of alkaline earth metal which is essentially covalent in nature : (A) So MgO (C) CaO (D) SrO 	$(A) \bigcirc 1.0 \times 10^{-12} M$
 (C) ○ 5.0 × 10⁻¹² M (D) ○ 4.0 × 10⁻¹³ M Question No.43 (Question Id - 37) Given below are two statements : one is labelled as Assertion A and the other is labelled as Reason R : Assertion A: Redox couple is the combination of oxidized and reduced form of a substance involved in an oxidation or reduction half cell. Reason R: In the representation E^O(Fe³⁺ /Fe²⁺) and E^O (Cu²⁺/Cu), (Fe³⁺/Fe²⁺) and (Cu²⁺/Cu) are redox couple. In the light of the above statements, choose the most appropriate answer from the options given below : (A) ○ Both A and R are correct but R is the correct explanation of A (B) ○ Both A and R are correct but R is NOT the correct explanation of A (Correct Answer) (C) ○ A is correct but R is correct (D) ○ A is not correct but R is correct (D) ○ A is not correct but R is correct (D) ○ A is not correct Answer) (C) ○ golgi apparatus (D) ○ cell wall Question No.43 (Question Id - 6) Identify the oxide of alkaline earth metal which is essentially covalent in nature : (A) ○ Both QUESTON Id - 80 (B) MgO (C) ○ CaO (D) ○ SrO 	(B) \bigcirc 2.0 x 10 ⁻¹³ M (Correct Answer)
 (D) ○ 4.0 × 10⁻¹³ M Cureation No.43 (Question Id - 37) Given below are two statements : one is labelled as Assertion A and the other is labelled as Reason R : Assertion A: Redox couple is the combination of oxidized and reduced form of a substance involved in an oxidation or reduction half cell. Reason R: In the representation E^O(Fe³⁺ /Fe²⁺) and E^O (Cu²⁺/Cu), (Fe³⁺/Fe²⁺) and (Cu²⁺/Cu) are redox couple. In the light of the above statements, choose the most appropriate answer from the options given below : (A) ○ Both A and R are correct and R is the correct explanation of A (B) ○ Both A and R are correct but R is NOT the correct explanation of A (B) ○ Both A and R are correct but R is NOT the correct explanation of A (Correct Answer) (C) ○ A is correct but R is not correct Cuestion No.44 (Question Id - 55) Inner membrane of which organelle forms infolding called cristae ? (A) ○ mitochondria (Correct Answer) (C) ○ golgi apparatus (D) ○ cell wall Cuestion No.45 (Question Id - 8) Identify the oxide of alkaline earth metal which is essentially covalent in nature : (A) ○ BeO (Correct Answer) (B) ○ MgO (C) ○ CaO (D) ○ SrO 	(C) \bigcirc 5.0 x 10 ⁻¹² M
Question No.43 (Question Id - 37) Given below are two statements : one is labelled as Assertion A and the other is labelled as Reason R : Assertion A: Redox couple is the combination of oxidized and reduced form of a substance involved in an oxidation or reduction half cell. Reason R: In the representation $E^{\Theta}(Fe^{3+}/Fe^{2+})$ and $E^{\Theta}(Cu^{2+}/Cu), (Fe^{3+}/Fe^{2+})$ and (Cu^{2+}/Cu) are redox couple. In the representation $E^{\Theta}(Fe^{3+}/Fe^{2+})$ and $E^{\Theta}(Cu^{2+}/Cu), (Fe^{3+}/Fe^{2+})$ and (Cu^{2+}/Cu) are redox couple. In the light of the above statements, choose the most appropriate answer from the options given below : (A) Both A and R are correct and R is the correct explanation of A (B) Both A and R are correct but R is NOT the correct explanation of A (Correct Answer) (C) A is correct but R is not correct (D) A is not correct but R is correct Question No.44 (Question Id - 55) Inner membrane of which organelle forms infolding called cristae ? (A) ribosomes (B) mich chondria (Correct Answer) (C) golg apparatus (D) cell wall Cuestion No.45 (Question Id - 6) Intermetry in a correct in nature : (A) BeO (Correct Answer) (B) MgO (C) CaO	(D) \bigcirc 4.0 x 10 ⁻¹³ M
Assertion A: Redox couple is the combination of oxidized and reduced form of a substance involved in an oxidation or reduction half cell. Reason R: In the representation E ^Θ (Fe ³⁺ /Fe ²⁺) and E ^Θ (Cu ²⁺ /Cu), (Fe ³⁺ /Fe ²⁺) and (Cu ²⁺ /Cu) are redox couple. In the light of the above statements, choose the most appropriate answer from the options given below : (A) ○ Both A and R are correct and R is the correct explanation of A (B) ○ Both A and R are correct but R is NOT the correct explanation of A (Correct Answer) (C) ○ A is not correct but R is not correct (D) ○ A is not correct but R is correct Question No.44 (Question Id - 55) Inner membrane of which organelle forms infolding called cristae ? (A) ○ ibosomes (B) ○ mitochondria (Correct Answer) (C) ○ golgi apparatus (D) ○ cell wall Question No.45 (Question Id - 8) Identify the oxide of alkaline earth metal which is essentially covalent in nature : (A) ○ Be0 (Correct Answer) (B) ○ Mg0 (C) ○ CaO (D) ○ SrO	Question No.43 (Question Id - 37) Given below are two statements : one is labelled as Assertion A and the other is labelled as Reason R :
Redox couple is the combination of oxidized and reduced form of a substance involved in an oxidation or reduction half cell. Reason R: In the representation E ^Θ (Fe ³⁺ /Fe ²⁺) and E ^Θ (Cu ²⁺ /Cu), (Fe ³⁺ /Fe ²⁺) and (Cu ²⁺ /Cu) are redox couple. In the light of the above statements, choose the most appropriate answer from the options given below : (A) ○ Both A and R are correct and R is the correct explanation of A (B) ○ Both A and R are correct but R is NOT the correct explanation of A (Correct Answer) (C) ○ A is correct but R is not correct (D) ○ A is not correct but R is correct Question No.44 (Question Id - 55) Inner membrane of which organelle forms infolding called cristae ? (A) ○ inbosomes (B) ○ mitochondria (Correct Answer) (C) ○ golgi apparatus (D) ○ cell wall Question No.45 (Question Id - 8) Identify the oxide of alkaline earth metal which is essentially covalent in nature : (A) ○ Be0 (Correct Answer) (B) ○ Mg0 (C) ○ CaO (B) ○ SrO	Assertion A:
Reason R: In the representation $E^{\Theta}(Fe^{3+}/Fe^{2+})$ and $E^{\Theta}(Cu^{2+}/Cu)$, (Fe^{3+}/Fe^{2+}) and (Cu^{2+}/Cu) are redox couple. In the light of the above statements, choose the most appropriate answer from the options given below : (A) O Both A and R are correct and R is the correct explanation of A (B) Both A and R are correct but R is NOT the correct explanation of A (Correct Answer) (C) A is correct but R is not correct (D) A is not correct but R is correct Question No.44 (Question Id - 55) Inner membrane of which organelle forms infolding called cristae ? (A) oritosomes (B) oritosomes (B) cell wall Question No.45 (Question Id - 6) Identify the oxide of alkaline earth metal which is essentially covalent in nature : (A) Be0 (Correct Answer) (B) Mg0 (C) CaO (D) SrO	Redox couple is the combination of oxidized and reduced form of a substance involved in an oxidation or reduction half cell.
In the representation $E^{\Theta}(Fe^{3^{+}}/Fe^{2^{+}})$ and $E^{\Theta}(Cu^{2^{+}}/Cu)$, $(Fe^{3^{+}}/Fe^{2^{+}})$ and $(Cu^{2^{+}}/Cu)$ are redox couple. In the light of the above statements, choose the most appropriate answer from the options given below : (A) \bigcirc Both A and R are correct and R is the correct explanation of A (B) \bigcirc Both A and R are correct but R is NOT the correct explanation of A (Correct Answer) (C) \bigcirc A is correct but R is not correct (D) \bigcirc A is not correct but R is correct Question No.44 (Question Id - 55) Inner membrane of which organelle forms infolding called cristae ? (A) \bigcirc ribosomes (B) \bigcirc mitochondria (Correct Answer) (C) \bigcirc golgi apparatus (D) \bigcirc cell wall Question No.45 (Question Id - 8) Identify the oxide of alkaline earth metal which is essentially covalent in nature : (A) \bigcirc BeO (Correct Answer) (B) \bigcirc MgO (C) \bigcirc CaO (D) \bigcirc SrO	Reason R:
In the light of the above statements, choose the most appropriate answer from the options given below : (A) ○ Both A and R are correct and R is the correct explanation of A (B) ○ Both A and R are correct but R is NOT the correct explanation of A (Correct Answer) (C) ○ A is correct but R is not correct (D) ○ A is not correct but R is correct Question No.44 (Question Id - 55) Inner membrane of which organelle forms infolding called cristae ? (A) ○ ribosomes (B) ○ mitochondria (Correct Answer) (C) ○ golgi apparatus (D) ○ cell wall Question No.45 (Question Id - 8) Identify the oxide of alkaline earth metal which is essentially covalent in nature : (A) ○ BeO (Correct Answer) (B) ○ MgO (C) ○ CaO (D) ○ SrO (C) ○ SrO	In the representation $E^{\Theta}(Ee^{3+}/Ee^{2+})$ and $E^{\Theta}(Cu^{2+}/Cu)$ (Ee^{3+}/Ee^{2+}) and (Cu^{2+}/Cu) are redox couple
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 (C) ○ golgi apparatus (D) ○ cell wall Question No.45 (Question Id - 8) Identify the oxide of alkaline earth metal which is essentially covalent in nature : (A) ○ BeO (Correct Answer) (B) ○ MgO (C) ○ CaO (D) ○ SrO 	(B) O mitochondria (Correct Answer)
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 (A) BeO (Correct Answer) (B) MgO (C) CaO (D) SrO 	Question No.45 (Question Id - 8)
(B) ○ MgO (C) ○ CaO (D) ○ SrO	(A) O BeO (Correct Answer)
(C) ○ CaO (D) ○ SrO	
(D) 🔿 SrO	$(C) \bigcirc CaO$
	(D) 🔿 SrO
Overstian No. 40 (Overstian 14, 44)	
Question No.46 (Question Id - 14) Acceptor type impurity in Si is formed by adding impurity of valency :	Question No.46 (Question Id - 14) Acceptor type impurity in Si is formed by adding impurity of valency :

(A) O 3 (Correct Answer) (B) 🔿 4 (C) 🔿 5 (D) 🔿 6 Question No.47 (Question Id - 40) $-\overset{|}{\underset{C}{\overset{}}}$ $\xrightarrow{}$ $C_{2}H_{5}O^{-}$ $\xrightarrow{}$ XH₃C -Where X is : (A) 🔿 OC_2H_5 $H_3C \longrightarrow CH_3$ H $\stackrel{(B)}{\overset{\bigcirc}{_{_{_{3}}}}} H_3C - \underset{H}{\overset{C}{_{_{_{2}}}}} C = CH - C_2H_5$ $H_3C - C = CH_2$ (Correct Answer) (C) 🔿 н - C ---- CH₃ (D) 🔿 H₃C -Question No.48 (Question Id - 42) When a photon stimulates the emission of another photon, the two photons have : A. Same direction B. Same energy C. Same phase D. Same wavelength Choose the correct answer from the options given below : (A) O A and C Only (B) O D and A Only (C) O C and B Only (D) O A, B, C and D (Correct Answer) Question No.49 (Question Id - 52) The internal energy of an ideal gas decreases by the same amount as the work done by the system. A. The process must be adiabatic B. The process must be isothermal C. The process must be isobaric D. The temperature must decrease Choose the correct answer from the options given below : (A) Only (B) O B Only (C) 🔿 C Only (D) O A and D Only (Correct Answer) Question No.50 (Question Id - 22) Vaccine against tuberculosis is called : (A) O Mycobacterium (B) O Bacille Calmette Guerin (BCG) (Correct Answer) (C) 🔘 S. typhi (D) O Francisellatularensis Question No.51 (Question Id - 35) pH of a solution of a strong acid is 5.0. The pH of the solution obtained after diluting the given solution a 100 times is (log 2 = 0.301)

(A) ○ 6.699 (Correct Answer)
 (B) ○ 7.001

(C) O 5.501

(D) 🔿 5.691
Question No.52 (Question Id - 45) A plane cuts intercepts a, 3b and 2c along the crystallographic axes in a crystal. The Miller indices of plane :
 (A) ○ (623) (Correct Answer) (B) ○ (263) (C) ○ (362) (D) ○ None of these
Question No.53 (Question Id - 47)
A normalized wave function is given by $\Psi = \frac{1}{\sqrt{3}} \varphi_0 + i \sqrt{\frac{2}{3}} \varphi_1$ where φ_0 and φ_1 are normalized Eigen functions with energies E_0 and E_1 , corresponding to ground state and first excited state respectively. What is the probability of getting E_0 when energy is measured ?
(A) \bigcirc 13 (correct Answer) (B) \bigcirc 1/ $\sqrt{3}$ (C) \bigcirc 1 (D) \bigcirc 0
Question No.54 (Question Id - 4) Which of the reaction is not a part of Sandmeyer reaction ? (A) \bigcirc ArN ₂ ⁺ X ⁻ $\xrightarrow{Cu_2Cl_2/HCl}$ ArCl + N ₂ (B) \bigcirc ArN ₂ ⁺ X ⁻ $\xrightarrow{Cu_2Cl_2/HBr}$ ArBr + N ₂ (C) \bigcirc ArN ₂ ⁺ X ⁻ $\xrightarrow{Cu/HCl}$ ArCl + N ₂ + CuX (Correct Answer) (D) \bigcirc ArN ₂ ⁺ X ⁻ $\xrightarrow{CuCN/KCN}$ ArCN + N ₂
Question No.55 (Question Id - 61)
Match the object in PART - A with their size in PART - B.PART - APART - BA. NanoshellI. 100 nmB. Hydrogen atomII. 2000 nmC. E. coli bacteriumIII. 90 nmD. TransistorIV. 0.1 nmChoose the correct answer from the options given below :(A) \bigcirc A - I, B - III, C - IV, D - II(B) \bigcirc A - I, B - I, C - III, D - IV(C) \bigcirc A - I, B - IV, C - II, D - III (Correct Answer)(D) \bigcirc A - III, B - IV, C - II, D - I
Question No.56 (Question Id - 7)For one component system, the phase rule is :(A) \bigcirc F = 3 - p (Correct Answer)(B) \bigcirc F = 2 - p(C) \bigcirc F = 1 - p(D) \bigcirc F = 1
Question No.57 (Question Id - 10) With the rise in temperature the conductance of a solution of an electrolyte generally : (A) O Decreases (B) Increases (Correct Answer) (C) Remain constant (D) A small decrease
Question No.58 (Question Id - 18) Let n _p and n _e be the number of holes and conduction electrons in an intrinsic semiconductor.
$(A) \bigcirc n_p > n_e$ $(B) \bigcirc n_p = n_e \text{ (Correct Answer)}$ $(C) \bigcirc n_p < n_e$ $(D) \bigcirc n_p \neq n_e$
Question No.59 (Question Id - 36)

Molar enthalpy change for vapourisation of 1 mol of water at 1 bar and 373 K is 41 kJ mol ⁻¹ . The internal energy change when 1 mol of water is vapourised is (if water vapour is assumed to be a perfect gas) :
(A) O 41 kJ mol ⁻¹
(B) O 37.9 kJ mol⁻¹ (Correct Answer)
(C) ○ - 41 kJ mol ⁻¹
(D) ○ 40.9 kJ mol ⁻¹
Question No.60 (Question Id - 57)
The most common stains used in Gram staining is :
(A) ○ crystal violet and inciripene blue (B) ○ crystal violet and safranin (Correct Answer)
(C) O crystal violet and carbol fuschin
(D) O safranin and methylene blue
Question No.61 (Question Id - 51) In an extrinsic semiconductor, in the region where mobility variation with temperature is evident, the slope of log (conductivity) vs. 1/T plot is :
(A) O negative
(B) O positive (Correct Answer)
(D) O infinite
Question No.62 (Question Id - 60)
Polio can lead to : (A) \bigcirc paralysis
(B) Onervous system distracted
(C) ⊖ both 1 and 2 (Correct Answer)
(D) ⊖ eye diseases
Question No.63 (Question Id - 62) Mode of DNA replication is : (A) Conservative and bidirectional (B) Semiconservative and unidirectional (C) Semiconservative and bidirectional (Correct Answer) (D) Conservative and unidirectional
Question No.64 (Question Id - 49) Magnetic field at a distance r from a long thin straight wire carrying current I is (μ_0 is susceptibility, ϵ_0 is permittivity of free space)
(A) ○ I/(4π∈₀r²)
(B) ○ I.r/€₀
(C) $\bigcirc l^2/(4\pi\epsilon_0 r)$
(D) Ο μ ₀ l/(2πr) (Correct Answer)
Question No.65 (Question Id - 12) When dielectric slab of dielectric constant K is inserted fully between plates of a capacitor, its capacitance :
(A) 🔿 increases by a factor of K (Correct Answer)
(B) O decreases by a factor of K
(C) ○ remains constant
(D) O none of these
SECTION 2 - Nano Electronics
Question No.1 (Question Id - 70) An 8-bits flash ADC is to be built. What is the minimum number of comparators required to build this circuit :
(A) 🔿 8
(B) O 63
(C) ○ 255 (Correct Answer) (D) ○ 256
Question No.2 (Question Id - 123)



(C) 🔿 13.7 mA

Question No.9 (Question Id - 117)

A Silicon PN junction diode under reverse bias has depletion region of width 10 μ m. The relative permittivity of Silicon, $\epsilon_r = 11.7$ and the permittivity of free space $\epsilon_o = 8.85 \times 10^{-12}$ Fm. The depletion capacitance of the diode per square meter is :

(A) 0 100.3 μF
(B) 0 10.35 μF (Correct Answer)
(C) 0 1.5 μF

(D) 🔿 20 µF

Question No.10 (Question Id - 116)

If $R_1 = R_2 = R_4 = R$ and $R_3 = 1.1R$ in the bridge circuit shown in the figure, then the reading in the ideal voltmeter connected between a and b is :



(B) ○ 1.138 V (C) ○ - 0.238 V (Correct Answer)

(D) 🔿 1.28 V

Question No.11 (Question Id - 110)

A master - slave flip flop has the characteristic that :

(A) $\bigcirc\;$ change in the output immediately reflected in the output

- (B) \bigcirc change in the output occurs when the state of the master is affected
- (C) \bigcirc change in the output occurs when the state of the slave is affected (Correct Answer)
- (D) \bigcirc both the master and the slave states are affected at the same time

Question No.12 (Question Id - 101)

A 4 bit ripple counter and a bit synchronous counter are made using flip flops having a propagation delay of 10 ns each. If the worst case delay in the ripple counter and the synchronous counter be R and S respectively, then :

 $\begin{array}{ll} (A) & \bigcirc & \mathsf{R} = 10 \; \text{ns}, \; \mathsf{S} = 40 \; \text{ns} \\ (B) & \bigcirc & \mathsf{R} = 40 \; \text{ns}, \; \mathsf{S} = 10 \; \text{ns} \; (\text{Correct Answer}) \\ (C) & \bigcirc & \mathsf{R} = 10 \; \text{ns}, \; \mathsf{S} = 30 \; \text{ns} \\ (D) & \bigcirc & \mathsf{R} = 30 \; \text{ns}, \; \mathsf{S} = 10 \; \text{ns} \end{array}$

Question No.13 (Question Id - 121)

The Boolean expression for the truth table shown is :

А	В	С	D
0	0	0	0
0	0	1	0
0	1	0	0
0	1	1	1
1	0	0	0
1	0	1	0
1	1	0	1
1	1	1	0

(A) $\bigcirc B(A + C)(\overline{A} + \overline{C})$ (Correct Answer) (B) $\bigcirc B(A + \overline{C})(\overline{A} + C)$ (C) $\bigcirc \overline{B}(A + \overline{C})(\overline{A} + C)$

 $(D) \bigcirc \overline{B}(A + C)(\overline{A} + \overline{C})$

 $10^{10}/\mbox{cm}^3$. The electron concentration is : (A) O zero (B) (B) 10¹⁰/cm³ (C) \bigcirc 10⁵/cm³ (Correct Answer) (D) O 1.5 x 10²⁵/cm³ Question No.15 (Question Id - 78) A single "bus" structure is primarily found in : (A) O main frame (B) \bigcirc super computers (C) O high performance machine (Correct Answer) (D) O mini and micro computers Question No.16 (Question Id - 97) Match items in List-I with items in List-II, most suitably. List-I List-II A. LED I. Heavy doping B. Avalanche photo diode II. Coherent radiation III. Spontaneous emission C. Tunnel diode D. LASER IV. Current gain Choose the correct answer from the options given below : (A) 🔘 A - I, B - II, C - IV, D - III (B) 🔘 A - II, B - III, C - I, D - IV (C) O A - III, B - IV, C - I, D - II (Correct Answer) (D) 🔿 A - II, B - I, C - IV, D - III Question No.17 (Question Id - 89) Calculate the gain of a negative feedback amplifier with an internal gain, A=100 and feedback factor β = 1/10. (A) 🔘 0.909 (B) 🔿 1.09 (C) O 9.09 (Correct Answer) (D) O 0.99 Question No.18 (Question Id - 109) Choose the correct one from among the alternatives A, B, C after matching an item from Group 1 most appropriate item in Group 2. Group 1 Group 2 A. Shift register I. Frequency division B. Counter II. Addressing in memory chips C. Decoder III. Serial to parallel data conversion Choose the correct answer from the options given below : (A) 🔘 A - III, B - II, C - I (B) O A - III, B - I, C - II (Correct Answer) (C) 🔿 A - II, B - I, C - III (D) 🔿 A - I, B - II, C - III Question No.19 (Question Id - 106) A series circuit R-L-C circuit has a Q of 100 and an impedance of (100 + j0) Ω at its resonant angular frequency of 10⁷ radian/sec. The value of R and L are respectively as : (A) 🔘 R = 100 Ω, L = 5 mH (B) 🔘 R = 50 Ω, L = 5 mH (C) \bigcirc R = 100 Ω , L = 1 mH (Correct Answer) (D) O R = 100 Ω, L = 15 mH

Question No.20 (Question Id - 105)

The turn ratio of transformer used in a bridge rectifier is $n_1 : n_2 = 12 : 1$. The primary was connected to 220 V, 50 Hz power mains. Assuming that the diode voltage drops to be zero, find the dc voltage (V_{dc}) across the load. Also find PIV of each diode.

 $V_{dc}(V)$ PIV (V)

 (A)
 8.24
 12.5

 (B)
 16.48
 25.9 (Correct Answer)

 (C)
 6.48
 5.9

 (D)
 32.48
 25.9

Question No.21 (Question Id - 112)

The Boolean function f implemented in the figure using two input multiplexes is :



Question No.22 (Question Id - 74)

What is a microprocessor ?

(A) \bigcirc A manually controlled device

- (B) \bigcirc Can be either manually controlled or program controlled
- (C) \bigcirc A program-controlled device (Correct Answer)
- (D) O None of these

Question No.23 (Question Id - 129)

Group I lists four different semiconductor devices. Match each device in Group - I with its characteristic property in Group - II :

Group - I	Group - II	
A. BJT	I. Population inversion	
B. MOS capacitor	II. Pinch-off voltage	
C. LASER diode	III. Early effect	
D. JFET	IV. Flat-band voltage	

Choose the correct answer from the options given below :

 Question No.24 (Question Id - 111)

 The bandgap of Silicon at 300 K is :

 (A) ○
 1.3 eV

 (B) ○
 0.7 eV

 (C) ○
 1.1 eV (Correct Answer)

 (D) ○
 1.4 eV

Question No.25 (Question Id - 66) n-type silicon is obtained by doping silicon with :

(A) O Germanium

(B) O Phosphorus (Correct Answer)

- (C) O Aluminium
- (D) O Boron

Question No.26 (Question Id - 124)



A particular green LED emits light of wavelength 5490 Angstron. The energy band-gap of the semiconductor material used there is : (Plank's constant = $6.626 \times 10^{-34} \text{ Js}$)
 (A) ○ 2.26 eV (Correct Answer) (B) ○ 1.98 eV (C) ○ 1.17 eV (D) ○ 0.74 eV
Question No.34 (Question Id - 68) To make the following statement correct, choose proper substitutes for X and Y : Tunnel diode and Avalanche photo diode are operated in X bias and Y bias respectively.
 (A) ○ X : reverse, Y : reverse (B) ○ X : reverse, Y : forward (C) ○ X : forward, Y : reverse (Correct Answer) (D) ○ X : forward, Y : forward
Question No.35 (Question Id - 87) The transistor is said to be a quiescent state when ? (A) no signal is applied to the input (Correct Answer) (B) it is unbiased (C) no current are flowing (D) emitter-junction bias is just equal to collector-junction bias
Question No.36 (Question Id - 80) What is the bandwidth between half power points for a circuit which resonates at 1MHz and has a Q of 100 ?
 (A) ○ 10 kHz (Correct Answer) (B) ○ 100 kHz (C) ○ 10 Hz (D) ○ 100 Hz
Question No.37 (Question Id - 81) The circuit analysis with nodal method is based on : (A) KVL and Ohm's Law (B) KCL and Ohm's Law (Correct Answer) (C) KCL and KVL (D) KCL, KVL and Ohm's Law
Question No.38 (Question Id - 67) $R = 2 k\Omega$, $L = 1 H$, and $C = 1/400 \mu F$ are present in a series RLC circuit. What is the resonant frequency ? (A) $\bigcirc 2 \times 10^4 Hz$ (B) $\bigcirc 10^4 Hz$ (C) $\bigcirc (1/\pi) 10^4 Hz$ (Correct Answer) (D) $\bigcirc 2\pi \times 10^4 Hz$
Question No.39 (Question Id - 127)
For the circuit shown in the figure, the Thevenin voltage and resistance looking into X - Y are : $ \begin{array}{c} 10\\ 10\\ 2i\\ 10\\ 2n\\ 0Y \end{array} $ (A) $\bigcirc 4/3 \vee, 2 \Omega$ (B) $\bigcirc 4/3 \vee, 2/3 \Omega$ (C) $\bigcirc 4/3 \vee, 2/3 \Omega$ (D) $\bigcirc 4 \vee, 2 \Omega$ (Correct Answer)
Question No.40 (Question Id - 84) A Zener diode : (A) has a high forward-voltage rating (B) has a sharp breakdown at low reverse voltage (Correct Answer) (C) is useful as an amplifier (D) has a negative resistance
Question No.41 (Question Id - 92) In the design of an analog circuit to avoid thermal run-away, the operating point of the BJT should be :

In the design of an analog circuit to avoid thermal run-away, the operating point of the BJT should be :

$(A) \bigcirc V_{CE} = (\frac{1}{2}) V_{CC}$
(B) \bigcirc V _{CE} \leq (½) V _{CC} (Correct Answer)
$(C) \bigcirc V_{CE} \ge (\frac{1}{2}) V_{CC}$
(D) \bigcirc V _{CE} = (0.78) V _{CC}
Question No.42 (Question Id - 93) The Boolean function Y = AB + CD is to be realized using only 2-input NAND gates. What is the minimum number of gates required to implement this function ?
(A) 🔿 2
(B) O 3 (Correct Answer)
$(C) \bigcirc 4$
(D) (D) 5
Question No.43 (Question Id - 100) An ideal sawtooth voltages waveform of frequency of 500 Hz and amplitude 3 V is generated by charging a capacitor of 2 μ F in every cycle. The charging requires :
(A) ○ Constant voltage source of 3 V for 1 ms
(B) ○ Constant voltage source of 3 V for 2 ms
(C) O Constant current source of 1 mA for 1 ms
(D) O Constant current source of 3 mA for 2 ms (Correct Answer)
Question No.44 (Question ld - 71)
Find the unit of $\Delta \mathbf{x} \mathbf{H}$.
(A) O Ampere
(B) O Ampere/meter ² (Correct Answer)
(C) Ampere/meter
(D) O Ampere-meter
Question No.45 (Question Id - 90) The current of bipolar transistor drops at higher frequencies because of : (A) Transistor capacitance (Correct Answer) (B) high current effect in base (C) parasitic inductance (D) the early effect
Question No.46 (Question Id - 107) A 0 to 6 counter consists of 3 flip flops and a combination circuit of 2 input gate (s). The common circuit consists of :
(A) 🔿 one AND gate
(B) ○ one OR gate
(C) O one AND gate and one OR gate
(D) O two AND gates (Correct Answer)
Question No.47 (Question Id - 108) A microprocessor with a 16-bit address bus is used in a linear memory selection configuration (i.e. Address bus lines are directly used as a chip select of memory chip) with 4 memory chips. The maximum addressable memory space is :
(A) Carrect Answer)
(A) ○ 64k (Correct Answer) (B) ○ 16k
 (A) ○ 64k (Correct Answer) (B) ○ 16k (C) ○ 8k
 (A) ○ 64k (Correct Answer) (B) ○ 16k (C) ○ 8k (D) ○ 4k
 (A) ○ 64k (Correct Answer) (B) ○ 16k (C) ○ 8k (D) ○ 4k Question No.48 (Question ld - 77)
 (A) ○ 64k (Correct Answer) (B) ○ 16k (C) ○ 8k (D) ○ 4k Question No.48 (Question Id - 77) Restart is a special type of CALL in which :
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In a uniformly doped BJT, assume that N_E , N_B and N_C are the emitter, base and collector doping in	
atoms/cm ³ , respectively. If the emitter injection efficiency of the BJT is close unity, which one of the following condition is TRUE ?	
$(A) \bigcirc N_{E} = N_{B}$	
(B) ○ N _E >> N _B (Correct Answer)	
$(C) \bigcirc N_E > N_B$	
$(D) \bigcirc N_{E} < N_{B}$	
Question No.51 (Question Id - 99) When the gate-to-source voltage (V _{GS}) of a MOSFET with threshold voltage of 400 mV, working in saturation is 900 mV, the drain current is observed to be 1 mA. Neglecting the channel width modulation effect and assuming that the MOSFET is operating at saturation, the drain current for an applied V _{GS} of 1400 mV is :	
(A) 🔿 0.5 mA	
(B) ○ 2.0 mA	
(C) \bigcirc 3.5 mA (D) \bigcirc 4.0 mA (Correct Answer)	
Question No.52 (Question Id - 91) The ideal op-amp has the following characteristics : (A) \bigcirc R _{in} = 0, A = ∞ and R _a = 0	
(B) \bigcirc R _{in} = ∞ , A = ∞ and R _o = 0 (Correct Answer)	
(C) \bigcirc R _{in} = 0, A = 0 and R _o = ∞	
(D) \bigcirc R _{in} = ∞ , A = 0 and R _o = ∞	
Question No.53 (Question Id - 125) The majority carriers in an n-type semiconductor have an average drift velocity v in a direction perpendicular to a uniform magnetic field B. The electric field E induced due to Hall effect acts in the direction :	
(A) ○ v x B	
(B) O B x v (Correct Answer)	
$(C) \bigcirc$ along v	
(D) O opposite to v	
(D) O opposite to v Question No.54 (Question Id - 126)	
(D) Opposite to v Question No.54 (Question Id - 126) Find the correct match between Group 1 and Group 2 :	
(D) ○ opposite to v Question No.54 (Question Id - 126) Find the correct match between Group 1 and Group 2 : Group 1 Group 2 A Variation diada L. Voltage reference	
(D) Opposite to v Question No.54 (Question Id - 126) Find the correct match between Group 1 and Group 2 : Group 1 Group 2 A. Varactor diode I. Voltage reference R. PIN diode II. High frequency switch	
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 (D) opposite to v Question No.54 (Question Id - 126) Find the correct match between Group 1 and Group 2 : Group 1 Group 2 A. Varactor diode I. Voltage reference B. PIN diode II. High frequency switch C. Zener diode III. Tuned circuits D. Schottky diode IV. Current controlled attenuator Choose the correct answer from the options given below : (A) △ A - IV, B - II, C - I, D - III (B) △ A - IV, B - II, C - I, D - III (B) △ A - II, B - IV, C - I, D - III (D) △ A - I, B - III, C - I, D - III (D) △ A - I, B - III, C - I, D - III (D) △ A - I, B - III, C - I, D - III (D) △ A - I, B - III, C - II, D - IV Question No.55 (Question Id - 114) A multistage amplifier consists of three stages. The voltage gains of the stages are 30, 50 and 80. Calculate the overall voltage gain in dB. (A) △ 100.3 dB (B) △ 100.9 dB (C) △ 101.88 dB (Correct Answer) (C) △ 101.88 dB (Correct Answer) 	
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 (D) opposite to v Question No.54 (Question Id - 126) Find the correct match between Group 1 and Group 2 : Group 1 Group 2 A. Varactor diode I. Voltage reference B. PIN diode II. High frequency switch C. Zener diode III. Tuned circuits D. Schottky diode IV. Current controlled attenuator Choose the correct answer from the options given below : (A) ○ A - IV, B - II, C - I, D - III (B) ○ A - II, B - IV, C - I, D - III (B) ○ A - II, B - IV, C - I, D - III (D) ○ A - I, B - III, C - II, D - III Question No.55 (Question Id - 114) A multistage amplifier consists of three stages. The voltage gains of the stages are 30, 50 and 80. Calculate the overall voltage gain in dB. (A) ○ 100.3 dB (B) ○ 100.9 dB (C) ○ 101.58 dB (Correct Answer) (D) ○ 160.0 dB Question No.56 (Question Id - 55) Class of amplifiers that operates with least distortion. (A) ○ Class B (B) ○ Class A (Correct Answer) 	
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